

EUROPEAN COMMISSION HEALTH & CONSUMER PROTECTION DIRECTORATE-GENERAL

Safety of the food chain **Chemicals, contaminants, pesticides** 

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# Guidance document on analytical quality control and validation procedures for pesticide residues analysis in food and feed.

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#### **Coordinators**:

Dr. Tuija Pihlström\_\_\_\_\_National Food Agency, Uppsala, Sweden

Dr. Amadeo R. Fernández-Alba \_\_\_\_\_ EURL-FV, University of Almería, Spain

#### Editors:

Mr. Stewart Reynolds	The Food and Environment Research Agency, York, United	
	Kingdom	
Mr. Richard J. Fussell	The Food and Environment Research Agency, York, United Kingdom	
Dr. André de Kok	NVWA - Netherlands Food and Consumer Product Safety Authority, Wageningen, The Netherlands	
Dr. Michelangelo Anastassiades EURL-SRM, CVUA Stuttgart, Fellbach, Germany		

## Advisory Group:

Dr. Mette Erecius Poulsen	EURL-CF, National Food Institute, Soeborg, Denmark	
Mr. Ralf Lippold	EURL-AO, CVUA, Freiburg, Germany	
Ms. Susanne Ekroth	National Food Agency, Uppsala, Sweden	
Dr. Miguel Gamón	EURL-FV, Generalitat Valenciana, Spain	
Mr. Octavio Malato	EURL-FV, University of Almería, Spain	
Ms. Carmen Ferrer Amate	EURL-FV, University of Almería, Spain	
Dr. Antonio Valverde	University of Almería, Spain	
Dr. Sonja Masselter	AGES GmbH, Competence Center for Residues of Plant Protection Products, Innsbruck, Austria	
Dr. Hans Mol	RIKILT, Wageningen, The Netherlands	
Dr. Magnus Jezussek	Bavarian Health and Food Safety Authority, Erlangen, Germany	
Dr. Darinka Stajnbaher	Institute of Public Health, Maribor, Slovenia	

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## Main changes introduced into the revision of Document N° SANCO/12495/2011 during the Updating Session held in Almeria, October 2013

- 1) Global reorganisation of the document structure (changing the order of many paragraphs).
- 2) Introduction of new parts and paragraphs mainly focused on informative and clarifications purposes:
  - a. In section B (Sampling, transport, traceability and storage of laboratory samples):
    - i. In *Storage* (Paragraph B6).
  - b. In section C (Sample analysis):
    - i. In **Sample preparation and processing**, *Pooling of samples* (Paragraphs C5 and C6).
    - ii. In Clean-up, concentration/reconstitution and storage of extracts (Paragraphs C8 and C9).
    - iii. In **Calibration for quantification**, subsection *General requirements*, (Paragraph C14).
    - iv. In **Calibration for quantification**, subsection *Standard addition* (Paragraphs C25 and C26).
    - v. In **Calibration for quantification,** subsection *Procedural standard calibration* (Paragraphs C29 and C30).
    - vi. In **Calibration for quantification,** subsection *Uses of various internal standards* (Paragraphs C32 C38).
  - c. In section G (Analytical method validation and performance criteria):
    - i. In Screening methods (Paragraph G6).
- 3) In the paragraph C15: The **drifting criteria** for the single injection of **bracketing calibration** standards has been generalised to **30 %** for all the concentrations (*in the* old version the criteria was between 20% and 30% depending on the concentration of the standards; #38).
- 4) In the paragraph C17: The criteria for accepting multi-level calibration has been generalised for all the concentrations: the individual residuals must not deviate by more than ±20% from the calibration curve in the relevant region (in the old version the criteria was 10% in cases where the MRL was approached or exceeded; #40).
- 5) In the paragraph C18: The criteria for the difference in the concentrations of two calibration levels to accept **calibration by interpolation between two levels** has been expanded to a **factor of 10** (*in the old version the criteria was a factor of 4; #39*).
- 6) In the paragraph C18: The criteria for accepting **calibration by interpolation between two levels** has been generalised for all the concentrations: the response of the standards at each level should **not differ by more than 20%**, taking the higher response as 100% (in the old version the criteria was 10% in cases where the MRL was approached or exceeded; # 39).
- 7) In the paragraph C19: The criteria for accepting **single-level calibration** has been changed and generalised for all the concentrations: the detector response of the analyte in the sample extract should be **within ±30%** of the single-level calibration standard response (*in the old version the criteria was within ±50%, or within ±20% where the MRL was exceeded; #41*).

- 8) In the paragraph D2: The **tolerance criteria for retention time** has been generalised to ±0.2 min for both GC and LC (in the old version the relative retention time tolerance criteria was ±0.5% for GC and ±2.5% for LC; #75).
- 9) In the paragraph D9: The default recommended as the maximum permitted tolerances for relative ion intensities using MS techniques (except for GC-EI-MS) has been changed and generalised to ±30% (see third column of Table 5) (in the old version the tolerance varies from 20-50% ;#80).
- 10) In the paragraph E1: A new sentence has been added about the expression of results of components with residue definition. "The results from the individual components analysed must be reported." (in the old version paragraph #82).
- 11) In the paragraphs F10 and F11 devoted to Testing and replacement of standards, a tolerance criteria has been included for the variability of multiple injections of the new and old solutions: RSD < 10% or RSD <15% with a minimum of 3 (in the old version the variability should be lower than 10%; #23).</p>
- 12) In the Annex A, the old Commodity Group 4 "High oil content" has been divided into two:

- Commodity Group 4a. "High oil content and very low water content."

- Commodity Group 4b. "High oil content and intermediate water content."

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## ANALYTICAL QUALITY CONTROL AND METHOD VALIDATION PROCEDURES FOR PESTICIDE RESIDUES ANALYSIS IN FOOD AND FEED

## A. Introduction and legal background

A1 The guidance in this document is intended for laboratories involved in official control of pesticide residues in food and feed in the European Union. The document describes the method validation and analytical quality control (AQC) requirements to support the validity of data reported within the framework of official controls on pesticide residues and used for checking compliance with maximum residue levels (MRLs), enforcement actions, or assessment of consumer exposure to pesticides.

The key objectives are:

- to provide a harmonized, cost-effective quality assurance and quality control system in the EU
- to ensure the quality and comparability of analytical results
- to ensure that acceptable accuracy is achieved
- to ensure that false positives or false negatives are avoided
- to support compliance with, and specific implementation of ISO/IEC 17025 (accreditation standard)

A2 This document is complementary and integral to the requirements in ISO/IEC 17025.

A3 The glossary (Appendix E) should be consulted for definitions and explanation of terms used in the text.

A4 In accordance with Article 12 of Regulation 882/2004, laboratories designated for official control of pesticide residues must be accredited to ISO/IEC 17025. According to Article 11 of Regulation 882/2004, analysis methods used in the context of official controls shall comply with relevant Community rules or with internationally recognised rules or protocols or, in the absence of the above, with other methods fit for the intended purpose or developed in accordance with scientific protocols. Where the above does not apply, validation of methods of analysis may further take place within a single laboratory according to an internationally accepted protocol.

According to Article 28 of Regulation 396/2005, technical guidelines dealing with the specific validation criteria and quality control procedures in relation to methods of analysis for the determination of pesticide residues may be adopted in accordance with the procedure referred to in Article 45(2) of this regulation. The present document entails mutually acceptable scientific rules for official pesticide residue analysis within the EU as agreed by all Member States of the European Union and constitutes a technical guideline in the sense of article 28 of Regulation 396/2005. It should thus be consulted during audits and accreditations of official pesticide residue laboratories according to ISO/IEC 17025.

## B. Sampling, transport, traceability and storage of laboratory samples

## Sampling

B1 Laboratory food samples should be taken in accordance with Directive 2002/63/EC or superseding legislation. For feed, the regulations are laid down in Appendix I of Regulation (EC) 152/2009 and amendments. Where it is impractical to take primary samples randomly within a lot, the method of sampling must be recorded.

## Transport

B2 Samples must be transported under appropriate conditions to the laboratory in clean containers and robust packaging. Polythene or polypropylene bags, ventilated if appropriate, are acceptable for most samples but low-permeability bags (e.g. nylon film) should be used for samples to be analysed for residues of fumigants. Samples of commodities pre-packed for retail sale should not be removed from their packaging before transport. Very fragile or perishable products (e.g. ripe raspberries) may have to be frozen to avoid spoilage and then transported in "dry ice" or similar, to avoid thawing in transit. Samples that are frozen at the time of collection must be transported without thawing. Samples that may be damaged by chilling (e.g. bananas) must be protected from both high and low temperatures.

B3 Rapid transport to the laboratory, preferably within one day, is essential for samples of most fresh products. The condition of samples delivered to the laboratory should approximate to that acceptable to a discerning purchaser, otherwise samples should normally be considered unfit for analysis.

## Traceability

B4 Samples must be identified clearly and indelibly, in a way to ensure traceability. The use of marker pens containing organic solvents should be avoided for labelling bags containing samples to be analysed for fumigant residues, especially if an electron capture detector is to be used.

B5 On receipt, each laboratory sample must be allocated a unique code by the laboratory.

## Storage

B6 Laboratory samples which are not analysed immediately should be stored under conditions that minimise decay. Fresh produce should be stored in the refrigerator, but typically no longer than 5 days. Dried products may be stored at room temperature, but if storage time is expected to exceed two weeks, they should be sub-sampled and stored in the freezer.

## C. Sample analysis

C1 All sample preparation and processing procedures should be undertaken within the shortest time practicable to minimise sample decay and pesticide losses. Analyses for residues of very labile or volatile pesticides should be started, and the procedures which could lead to loss of analyte should be completed as soon as possible, but preferably on the day of sample receipt.

## Sample preparation and processing

C2 Sample preparation, sample processing and sub-sampling to obtain analytical portions should take place before visible deterioration occurs. The parts of the commodity that should be analysed are stipulated in Regulation 396/2005 Annex1.

C3 Sample processing and storage procedures should be demonstrated to have no significant effect on the residues present in the analytical sample (see Directive 2002/63/EC). Where there is evidence that comminution (cutting and homogenisation) at ambient temperature has a significant influence on the degradation of certain pesticide residues, it is recommended that samples are homogenised at low temperature (e.g. frozen and/or in the presence of "dry ice"). Where comminution is known to affect residues (e.g. dithiocarbamates or fumigants) and practical alternative procedures are not available, the test portion should consist of whole units of the commodity, or segments removed from large units. For all other analyses, the whole laboratory sample needs to be comminuted. To improve the extraction efficiency of low moisture containing commodities (e.g. cereals, spices, dried herbs), it is recommended that small particle sizes, preferably less than 1 mm, are obtained. Milling should be performed in a way that avoids extensive heating of the samples, as heating can cause losses of certain pesticides.

C4 Sample comminution should ensure that the sample is homogeneous enough so that sub-sampling variability is acceptable. If this is not achievable, the use of larger test portions or replicate portions should be considered to obtain a better estimate of the true value. Upon homogenization or milling, samples may separate into different fractions, e.g. pulp and peel in the case of fruits, and husks and endosperm in the case of cereals. This fractionation can occur because of differences in size, shape and density. Because pesticides can be heterogeneously distributed between the different fractions, it is important to ensure that the fractions in the analytical test portion are in the same ratio as in the original laboratory sample. It is advisable to store in a freezer a sufficient number of sub-samples or analytical test portions for the number of analyses/repeat analyses likely to be required.

## Pooling of samples

C5 Pooling of individual samples or sample extracts may be considered as an option for the analyses of commodities with a low frequency of pesticide residues (e.g. organic or animal products), provided that the detection system is sensitive enough. For example, when pooling 5 samples, the LOQ or SDL must be at least 5 times lower than the RL.

C6 Pooling of sub-samples before extraction will reduce the number of analyses required, but in some cases additional mixing or homogenisation of the pooled sub-samples, before withdrawing the analytical portion, may be necessary. Alternatively, sample extracts can be pooled before injection. The original samples or the extracts must be re-analysed in case of pesticide residue findings at relevant levels.

## Extraction

## Extraction conditions and efficiency

C7 The percentage recovery of incurred residues can be lower than the percentage recovery obtained from the analysis of spiked samples. Where practicable, samples containing incurred residues can be analysed at varying extraction conditions to obtain further information on extraction efficiency. A number of parameters such as temperature, pH, time etc., can affect extraction efficiency and analyte stability. To improve the extraction efficiency of low moisture containing commodities (cereals, dried fruits), addition water to the samples prior to extraction is recommended. The impact of shaking time on analyte losses should be checked to avoid unacceptable losses. Where the MRL residue definition of a pesticide includes salts, it is important that the salts are dissociated by the analytical procedure used. This is typically achieved by the addition of water before, or during, the extraction process. A change of pH may also be necessary. Where the residue definition includes esters or conjugates that cannot be analysed directly the analytical procedure should involve a hydrolysis step.

## Clean-up, concentration/reconstitution and storage of extracts

C8 A clean-up, enrichment or dilution step may be necessary to reduce matrix interferences resulting in improved selectivity, and to reduce contamination of the instrument system leading to improved robustness. Clean-up techniques take advantage of the difference of physicochemical properties (e.g. polarity, solubility, molecular size) between the pesticides and matrix components. However, the use of a clean-up step in a multi-residue method can result in losses of some pesticides.

C9 It should be considered that concentration of sample extracts can cause precipitation of matrix-components and in some cases result in losses of pesticides. Dilution of the extract with a solvent of a different polarity can also result in pesticide losses because of decreased solubility (e.g. dilution of methanol or acetonitrile with water).

C10 To avoid losses during evaporation steps the temperature should be kept as low as practicable. A small volume of high boiling point solvent may be used as a "keeper". Also, foaming and vigorous boiling of extracts, or dispersion of droplets, must be avoided. A stream of dry nitrogen or vacuum centrifugal evaporation is generally preferable to the use of an air stream for small-scale evaporation, as air is more likely to lead to oxidation or to introduce water and other contaminants.

C11 Analyte stability in extracts should be evaluated during method validation. Storage of extracts in a refrigerator or freezer will minimise degradation. Losses of pesticides in extracts at room temperature can occur, e.g. in vials in an auto sampler rack.

## Chromatographic separation and determination

C12 Sample extracts are normally analysed using capillary gas chromatography (GC) and liquid chromatography (LC) coupled to mass spectrometry (MS) for the identification and quantification of pesticides in food and feed samples. Various MS detection systems can be used, such as single or triple quadrupole, ion trap, time of flight, orbitrap. Typical ionisation techniques are: electron impact (EI), chemical ionisation (CI), atmospheric pressure chemical ionisation (APCI) and electrospray ionisation (ESI). Different acquisition modes may be used such as full-scan, selected ion monitoring (SIM), selected reaction monitoring (SRM) and multiple reaction monitoring (MRM).

C13 Nowadays, selective detectors for GC (ECD, FPD, PFPD, NPD) and LC (DAD, fluorescence) are less widely used as they offer only limited specificity. Their use, even in

combination with different polarity columns, does not provide unambiguous identification. These limitations may be acceptable for frequently found pesticides, especially if some results are also confirmed using a more specific detection technique. Such limitations in the degree of identification should be acknowledged when reporting the results.

## Calibration for quantification

## General requirements

C14 The Lowest Calibration Level (LCL) must be equal to or lower than the calibration level corresponding to the Reporting Limit (RL). The RL must not be lower than the LOQ.

C15 Bracketing calibration must be used unless the determination system has been shown to be free from significant drift, e.g. by monitoring the response of an internal standard. The calibration standards should be injected at least at the start and end of a sample sequence. If the drift between two bracketing injections of the same calibration standard exceeds 30% the bracketed samples containing pesticide residues should be re-analysed. Results for those samples that do not contain any of those analytes showing unacceptable drift can be accepted provided that the response at the calibration level corresponding to the reporting limit (RL) remained measurable throughout the batch, to minimise the possibility of false negatives. If required, priming of the GC or LC system should be performed immediately prior to the first series of calibration determinations in a batch of analyses.

C16 The detector response from the pesticides in the sample extract should lie within the range of the responses from the calibration standard solutions injected. Where necessary extracts containing high-level residues above the calibrated range must be diluted. If the calibration solutions are matrix-matched (paragraph C22) the matrix concentration in the calibration standard should also be diluted proportionately.

C17 Multi-level calibration (three or more levels) is preferred. An appropriate calibration function must be used and the calibration curve should not normally be forced through the origin without justification. The fit of the calibration function must be plotted and inspected visually and/or by calculation of the residuals, avoiding over-reliance on correlation coefficients, to ensure that the fit is satisfactory within the concentration range of the pesticides detected. If individual residuals deviate by more than  $\pm 20\%$  from the calibration curve in the relevant region, an alternative calibration function must be used. In general, the use of weighted linear regression (1/x) is recommended, rather than linear regression.

C18 Calibration by interpolation between two levels is acceptable providing the difference between the 2 levels is not greater than a factor of 10 and providing the response factors of the bracketing calibration standards are within acceptable limits. The response factor of bracketing calibration standards at each level should not differ by more than 20% (taking the higher response as 100%).

C19 Single-level calibration may also provide accurate results if the detector response of the analyte in the sample extract is close to the response of the single-level calibration standard (within ±30%). Where an analyte is spiked to a sample for recovery determination at a level corresponding to the LCL, recovery values <100% may be calculated using a single point calibration at the LCL. This particular calculation is intended only to indicate analytical performance achieved at the LCL and does not imply that residues <LCL should be determined in this way.

## Representative analytes for calibration

C20 Where practicable, all targeted analytes should be injected in every batch of samples, at least at the level corresponding to the RL. Sufficient response at this level is required and should be checked to avoid false negatives. If this requires a disproportionate effort, the

determination system must be calibrated with a minimum number of representative analytes. Reliance on representative pesticides only increases the risk of producing false negative results for non-represented pesticides. The choice of the representative analytes should take into account the pesticides most likely to be found in the samples to be analysed, as well as the physico-chemical characteristics of those pesticide that are difficult to analyse (analytes likely to give the poorest and most variable response). The number of representative analytes to be calibrated in each batch must be at least 15, plus 25% of the total number of analytes included in the scope of each instrument method. For example, if the analytical scope of an instrument method covers 40 analytes, the determination system must be calibrated with at least 25 representative analytes. If the scope of analysis in the determination system is 20 or less, then all analytes should be calibrated. The minimum frequency for calibration of representative analytes is given in Table 1.

Table	1. Minimum	frequency	of calibration
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	Representative analytes	All other analytes
Minimum frequency of calibration	In each batch of analyses.	Within a rolling programme at least every third month*
	At least one calibration level corresponding to the reporting limit.	At least one calibration level corresponding to the reporting limit See also paragraph C21

\* The minimum requirements are

(i) at the beginning and end of a survey or programme and

(ii) when potentially significant changes are made to the method.

C21 Where an analyte that is not a representative analyte is detected in a sample, at or above the RL, the sample must be re-analysed, using a quantitative method. When the tentative result indicates that a MRL might be exceeded, the sample must be re-analysed and accompanied by acceptable recovery of the detected analyte. The recovery test may be omitted when the standard addition approach is used or when using the isotope-dilution approach with the isotope-labelled internal standard being added to the analytical portion prior to extraction.

## Matrix-matched calibration

C22 Matrix effects are known to occur frequently in both GC and LC methods and should be assessed at the initial method validation stage. Matrix-matched calibration is commonly used to compensate for matrix effects. Extracts of blank matrix, preferably of the same type as the sample, should be used for calibration. An alternative practical approach to compensate for matrix effects in GC-analyses is the use of analyte protectants that are added to both the sample extracts and the calibration solutions in order to equalise the response of pesticides in solvent calibrants and sample extracts. The most effective way to compensate for matrix effects is the use of standard addition or use of isotopically labeled internal standards.

C23 In GC, representative matrix calibration, using a single representative matrix or a mixture of matrices, can be used to calibrate a batch of samples containing different commodities. Although this is preferable to the use of calibration standards in solvent, compared to exact matrix matching, it is likely that the calibration will be less accurate. It is recommended that the relative matrix effects are assessed and the approach modified accordingly.

C24 Compensation of matrix effects in LC-MS is more difficult to achieve because the matrix effects depend on the co-elution of each individual pesticide with co-extracted matrix components, which vary between different commodities. The use of matrix-matched calibration is, therefore, likely to be less effective compared to GC.

## Standard addition

C25 Standard addition is an alternative approach to the use of matrix-matched calibration standards. This procedure is designed to compensate for matrix effects and recovery losses, but not extraction efficiency or chromatographic interferences caused by overlapping/unresolved peaks from co-extracted compounds. This technique assumes some knowledge of the likely residue level of the analyte in the sample (e.g. from a first analysis), so that the amount of added analyte is similar to that already present in the sample. In particular, it is recommended that standard addition is used for confirmatory quantitative analyses in cases of MRL exceedances and/or when no suitable blank commodity is available for the preparation of matrix-matched standard solutions. For standard addition a test sample is divided in three (or preferably more) test portions. One portion is analysed directly, and increasing amounts of the standard analyte are added to the other test portions immediately prior to extraction. The amount of the standard analyte added to the test portion should be between one and five times the estimated amount of the analyte already present in the sample. The concentration of analyte present in the "unspiked" sample extract is calculated from the relative responses of the analyte in the sample extract and the spiked samples extracts. In the standard addition approach the concentration of the analyte in the test sample extract is derived by extrapolation, thus a linear response in the appropriate concentration range is essential for achieving accurate results.

C26 Addition of at least two known quantities of analyte to aliquots of the sample extract, e.g. prior to injection, is another form of standard addition, but in this case adjustment is only for possible injection errors and matrix effects, but not for recovery losses.

## Effects of pesticide mixtures on calibration

C27 The detector response of individual pesticides in multi-pesticide calibration standards may be affected by one or more of the other pesticides in the same solution. Before use, multi-pesticide calibration solutions prepared in pure solvent should be checked against calibration standard solutions each containing a single pesticide (or a fewer number of pesticides) to confirm similarity of detector response. If the responses differ significantly, residues must be quantified using individual calibration standards in matrix, or better still, by standard addition.

## Calibration for pesticides that are mixtures of isomers

C28 Quantification involving mixed isomer (or similar) calibration standard solutions, can be achieved by using either: summed peak areas, summed peak heights, or measurement of a single component, whichever is the most accurate.

## Procedural Standard Calibration

C29 The use of procedural standards is an alternative type of calibration. This approach can compensate for matrix effects and low extraction recoveries associated with certain pesticide/commodity combinations, especially where isotopically labeled standards are not available or too costly. It is only applicable when a series of samples of the same type are to be processed within the same batch (e.g. products of animal origin, products with high fat content). Procedural standards are prepared by spiking a series of blank test portions with different amounts of analyte, prior to extraction. The procedural standards are then analysed in exactly the same way as the samples.

C30 Another type of application of procedural standard calibration is where pesticides need to be derivatised, but reference standards of the derivatives are not available or the derivatisation yield is low or highly matrix dependent. In such cases it is recommended to spike the standards to blank matrix extracts just prior to the derivatisation step. In this case the procedural standard calibration will also compensate for varying derivatisation yields.

## Calibration using derivative standards or degradation products

C31 Where the pesticide is determined as a derivative or a degradation product, the calibration solutions should be prepared from a "pure" reference standard of the derivative or degradation product, if available. Procedural standards should only be used if they are the only practical option.

## Use of various internal standards

C32 An internal standard (IS) is a chemical compound added to the sample test portion or sample extract in a known quantity, at a specified stage of the analysis, in order to check the correct execution of (part of) the analytical procedure. The IS should be chemically stable and/or typically show the same behaviour as of the target analyte.

C33 Depending on the stage of the analytical procedure in which the addition of IS takes place different terms are used. An injection internal standard (I-IS), also called instrument internal standard, is added to the final extracts, just prior to the determination step (i.e. at injection). It will allow a check and possible correction for variations in the injection volume. A procedural internal standard (P-IS) is an internal standard added at the beginning of the analytical procedure to account for various sources of errors throughout all stages in the method. The IS can also be added at a different stage of the analytical procedure to correct for both systematic and random errors that may have occurred during a specific stage of the analytical procedure. When selecting ISs it should be assured that they do not interfere with the analysis of the target analytes and that it is highly unlikely that they are present in the samples to be analysed.

C34 For multi-analyte methods it is advisable to use more than one IS in case the recovery or detection of the primary IS is compromised. If only used to adjust for simple volumetric variations the ISs should exhibit minimal losses or matrix effects. When dealing with a specific group of analytes with similar properties the IS can be chosen to exhibit similar properties and analytical behaviour to the compounds of interest. If the IS used for calculations has a significantly different behaviour (e.g. as to recovery or matrix effect) to one or more of the target analytes it will introduce an additional error in all quantifications.

C35 When the IS is added to each of the calibration solutions in a known concentration the detector response ratio of analyte and IS obtained from the injected calibration solutions is then plotted against the respective concentrations. The concentration of analyte is then obtained by comparing the detector response ratio of analyte and IS of the sample extract, against the calibration curve.

C36 An isotopically labeled internal standard (IL-IS) is an internal standard with the same chemical structure and elemental composition as the target analyte, but one or more of the atoms of the molecule of the target analyte are substituted by isotopes (e.g. deuterium, <sup>15</sup>N, <sup>13</sup>C, <sup>18</sup>O). A prerequisite for the use of IL-ISs is the use of mass spectrometry, which allows the simultaneous detection of the co-eluting non-labeled analytes and the corresponding IL-ISs. IL-ISs accurately compensate for both analyte losses and volumetric variations during the procedure, as well as for matrix effects and response drift in the chromatography-detection system. Losses during extract storage (e.g. due to degradation) will also be corrected for by the IL-IS. Use of IL-ISs will not compensate for incomplete extraction of incurred residues.

C37 Isotopically labeled internal standards (IL-ISs), can also be used to facilitate the identification of analytes because the retention time and peak shape of the target analyte and corresponding IL-IS should be the same.

C38 IL-ISs should be largely free of the respective native compounds to minimize the risk of false positive results. In the case of deuterated standards, an exchange of deuterium with

hydrogen atoms, e.g. in solvents, can lead to false positives and/or adversely influence quantitative results.

## Data processing

C39 Chromatograms must be examined by the analyst and the baseline fit checked and adjusted, as is necessary. Where interfering or tailing peaks are present, a consistent approach must be adopted for the positioning of the baseline. Peak area or peak height, whichever yields the more accurate results, may be used.

## On-going method performance verification during routine analysis

#### Quantitative methods

#### Routine recovery check

C40 Where practicable, recoveries of all target analytes should be measured within each batch of analyses. If this requires a disproportionately large number of recovery determinations, the number of analytes may be reduced, however, it should be in compliance with the minimum number specified in Table 2. This means, that at least 10% of the representative analytes (with a minimum of 5) should be included per detection system.

 Table 2. Minimum frequency of recovery checks (quantitative method performance verification)

	Representative analytes	All other analytes
Minimum frequency	10% of representative analytes (at least 5) per detection system, in each batch of	Within a rolling programme to include all other analytes at least every 12
of	analyses	months, but preferably every 6 months
recovery	Within a rolling programme covering all	At least at the level corresponding to
checks	representative analytes as well as representative commodities from different commodity groups, at least at the level corresponding to the reporting limit	the reporting limit

C41 If at some point during the rolling programme (Table 2) the recovery of an analyte is outside of the acceptable range (see paragraph C45), then all of the results produced since the last satisfactory recovery must be considered to be potentially erroneous.

C42 The recovery of an analyte should normally be determined by spiking within a range corresponding to the RL and 2-10 x the RL, or at the MRL, or at a level of particular relevance to the samples being analysed. The spiking level may be changed to provide information on analytical performance over a range of concentrations. Recovery at levels corresponding to the RL and MRL is particularly important. In cases where blank material is not available (e.g. where inorganic bromide is to be determined at low levels) or where the only available blank material contains an interfering compound, the spiking level for recovery should be  $\geq$ 3 times the level present in the blank material. The analyte (or apparent analyte) concentration in such a blank matrix extract should be determined from multiple test portions. If necessary, recoveries can be calculated using blank subtracted calibration, but the use of blank subtraction should be reported with the results. They must be determined from the matrix used in spiking experiments and the blank values should not be higher than 30% of the residue level corresponding to the RL.

C43 Where a residue is determined as a common moiety, routine recovery may be determined using the component that either normally predominates in residues or is likely to provide the lowest recovery.

C44 Where results are expressed on the basis of fat content or dry weight, the method used to determine the dry weight or fat content must be consistent. Ideally it should be validated against a widely recognised method. For feeding stuffs the methods listed in Appendix III of Directive (EC) No 152/2009 are obligatory.

## Acceptance criteria for routine recoveries

C45 Acceptable limits for individual recovery results should normally be within the range of the mean recovery +/- 2x RSD. For each commodity group (see Annex A) the mean recovery results and RSDs may be taken from initial method validation or from on-going recovery results (within laboratory reproducibility, RSD<sub>R</sub>). A practical default range of 60-140 % may be used for individual recoveries in routine multi-residue analysis. Recoveries outside the above mentioned range would normally require re-analysis of the batch, but the results may be acceptable in certain justified cases. For example, where the individual recovery is unacceptably high and no residues are detected, it is not necessary to re-analyse the samples to prove the absence of residues. However, consistently high recoveries or RSDs outside  $\pm 20\%$  must be investigated.

C46 Analysis of certified reference materials (CRMs) is the preferable option to provide evidence of method performance. However, CRMs that contain the relevant analytes at appropriate levels are seldom available. As an alternative, in-house reference materials may be analysed regularly instead. Where practicable, exchange of such materials between laboratories provides an additional, independent check of accuracy.

## Screening methods

C47 For qualitative multi-residue methods targeting very large numbers of analytes, it may not be practicable to include all analytes from the scope in each batch of analyses. To verify overall method performance for each batch, at least 10 representative (indicator) analytes (from the validated scope) that cover all critical points of the method should be spiked to a matrix. In a rolling programme, the performance for all analytes from the validated scope should be verified as indicated in Table 3.

	Representative (indicator) analytes	All other analytes
Number of analytes	At least 10 analytes per detection system covering all critical aspects of the method	All analytes from the validated qualitative scope
Minimum frequency of recovery checks	Every batch	At least every 12 months, preferably every 6 months
Level	SDL	SDL
Criterion	All (indicator) analytes detectable	All (validated) analytes detectable

 Table 3. Minimum frequency of recovery checks (screening method performance verification).

## Proficiency testing

C48 For all official control laboratories it is mandatory to participate regularly in proficiency test schemes, particularly those organised by the EURLs. When false positive(s) or negative(s) are reported, or the accuracy (z-scores) achieved in any of the proficiency tests is questionable or unacceptable, the problem(s) should be investigated. False positive(s), negative(s) and, or unacceptable performance, have to be rectified before proceeding with further determinations of the analyte/matrix combinations involved.

## D. Identification of analytes and confirmation of results

## Identification

## Mass spectrometry coupled to chromatography

D1 Mass spectrometry coupled to a chromatographic separation method is a very powerful combination for identification of an analyte in the sample extract. It simultaneously provides retention time, ion/charge ratios and relative abundance (intensity) data.

## Requirements for chromatography

D2 The minimum acceptable retention time for the analyte(s) under examination should be at least twice the retention time corresponding to the void volume of the column. The retention time of the analyte in the extract should correspond to that of the calibration standard (may need to be matrix-matched) with a tolerance of ±0.2 min, for both gas chromatography and liquid chromatography. Greater retention time deviations are acceptable where both retention time and peak shape of the analyte match with those of a suitable IL-IS, or evidence from validation studies is available. IL-IS can be particularly useful where the chromatographic procedure exhibits matrix-dependent retention time shifts or peak shape distortions. Overspiking with the analyte suspected to be present in the sample will also help to increase confidence in the identification.

## Requirements for mass spectrometry (MS)

D3 Reference spectra for the analyte should be generated using the same instruments and techniques used for analysis of the samples. If major differences are evident between a published spectrum and the spectrum generated within the laboratory, the latter must be shown to be valid. To avoid distortion of ion ratios the response of the analyte ions must not overload the detector. The reference spectrum in the instrument software can originate from a previous injection (without matrix present), but preferably obtained from the same analysis batch.

D4 Identification relies on proper selection of diagnostic (characteristic) ions. The (quasi) molecular ion is a diagnostic ion that should be included in the measurement and identification procedure whenever possible. In general, and especially in single-stage MS, high m/z ions are more specific than low m/z ions (e.g. m/z < 100). However, high mass m/z ions arising from loss of water or loss of common moieties may be of little use. Although characteristic isotopic ions, especially CI or Br clusters, may be particularly useful, the selected diagnostic ions should not exclusively originate from the same part of the parent molecule. The choice of diagnostic ions may change depending on background interferences.

D5 Extracted ion chromatograms of sample extracts should have peaks (exceeding S/N 3:1) of similar retention time, peak shape and response ratio to those obtained from a calibration standard analysed at comparable concentration in the same batch. Chromatographic peaks from different selective ions for the same analyte must overlap with each other. Where an ion chromatogram shows evidence of significant chromatographic interference, it must not be relied upon to quantify or identify residues. The ion that shows the best signal-to-noise ratio and no evidence of significant chromatographic interference should be used for quantification.

D6 In case of full scan measurement, careful subtraction of background spectra, either manual or automatic, by deconvolution or other algorithms, may be required to ensure that the resultant spectrum of the chromatographic peak is representative. Whenever background correction is used, this must be applied uniformly throughout the batch and should be clearly indicated.

D7 Different types and modes of mass spectrometric detectors provide different degrees of selectivity and specificity, which relates to the confidence in identification. The requirements for identification are given in Table 4. They should be regarded as guidance criteria for identification, not as absolute criteria to prove presence or absence of a compound.

MS mode:	Single-stage MS (unit mass resolution)	Single-stage MS (high resolution/high mass accuracy)	MS/MS
Typical systems (examples):	Quadrupole, ion trap, time-of- flight (TOF)	TOF, Orbitrap, FTMS, magnetic sector	Triple quadrupole, ion trap, hybrid MS (e.g. Q-TOF, Q-trap)
Acquisition mode:	Full scan, Limited m/z range, Selected ion monitoring (SIM)	Full scan, Limited m/z range, Selected ion monitoring (SIM)	Selected/multiple reaction monitoring (SRM/MRM),full scan product-ion spectra
Requirements for identification:	≥ 3 diagnostic ions, preferably including the (quasi) molecular ion	≥ 2 diagnostic ions, preferably including the (quasi) molecular ion; mass accuracy < 5 ppm; at least one fragment ion	≥ 2 product ions
	Ion ratio(s): according to Table 5		

Table 4. Identification	n criteria for diff	erent MS techniques
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D8 The relative intensities or ratios of selective ions (full-scan MS or SIM) or product ions (MS/MS), expressed as a ratio relative to the most intense (product) ion, should correspond to those of the calibration standard at comparable concentrations and measured under the same conditions. Matrix-matched calibration solutions may need to be used. Table 5 below indicates the recommended maximum tolerances for ion ratios.

D9 The variability of ion ratios should preferably be determined from calibration standards during initial method validation and subsequently during routine analysis. In certain cases, these data may be used to set performance-based criteria, for individual analytes, rather than applying the fixed, generic criteria given in Table 5.

Table 5. Recommended maximum (c	default) tolerances for ion ratio	s using different MS techniques
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Ion ratio (least/most intense ion)	Maximum tolerance (relative) for GC-EI-MS	Maximum tolerance (relative) for LC-MS <sup>n</sup> , LC-MS, GC-MS <sup>n</sup> , GC-CI-MS
0.50-1.00	± 10 %	± 30 %
0.20-0.50	± 15 %	± 30 %
0.10-0.20	± 20 %	± 30 %
<0.10	± 50 %	± 30 %

D10 Larger tolerances may lead to a higher percentage of false positive results. Similarly, if the tolerances are decreased, then the likelihood of false negatives will increase. The tolerances given in Table 5 should not be taken as absolute limits and automated data interpretation based on the criteria without complementary interpretation by an experienced analyst is not recommended.

D11 For a higher degree of confidence in identification, further evidence may be achieved from additional mass spectrometric information. For example, evaluation of full scan spectra, isotope pattern, adduct ions, additional accurate mass fragment ions, additional product ions (in MS/MS), or accurate mass product ions.

D12 The chromatographic profile of the isomers of an analyte may also provide evidence. Additional evidence may be sought using a different chromatographic separation system and/or a different MS-ionisation technique.

## Confirmation of results

D13 If the initial analysis does not provide unambiguous identification or does not meet the requirements for quantitative analysis, a confirmatory analysis is required. This may involve reanalysis of the extract or the sample. In cases where a MRL is exceeded, a confirmatory analysis of another portion of the homogenised laboratory sample is always required. For unusual pesticide/matrix combinations, a confirmatory analysis is also recommended.

D14 The use of different determination techniques and/or confirmation of qualitative and/or quantitative results by an independent expert laboratory will provide further supporting evidence.

## E. Reporting results

## Expression of results

E1 Results for individual analytes should be expressed as defined by the MRL residue definition and in mg/kg for food. For feeding stuffs, results should be expressed in mg/kg on a dry weight basis, assuming a moisture content of 12%. Where the residue definition includes more than one component (for examples, see Appendix B), the respective sum of components must be calculated as stated in the residue definition and must be used for checking compliance with the MRL. If the analytical capabilities of a laboratory do not allow quantification of the full sum of a residue as stated in the residue definition, a part of the sum may be calculated but this should be clearly indicated in the report. The results from the individual components analysed must always be reported.

E2 For quantitative methods, residues for individual analytes below the RL must be reported as <RL mg/kg. Where screening methods are used and a pesticide is not detected, the result must be reported as <SDL mg/kg.

E3 Where a residue definition is expressed as a sum of two or more measurable components the RL should be reported for each measurable component separately. Where a residue definition is expressed as a sum of various components that are quantified as a common moiety the RL should be derived by applying the entire method. Where required the RL should be expressed as stated in the residue definition by applying the appropriate conversion factor (Appendix B). Where the residue definition is expressed as a sum of isomers that are quantified as a sum (e.g. following integration of a group of peaks) the RL should apply to the sum of isomers.

## Calculation of results

E4 Where a residue has been identified and quantified using a single test portion, and the residue does not exceed the MRL, the reported result should be that measured using the detection technique considered to be the most accurate. Where results are obtained by two or more equally accurate techniques, the mean value may be reported.

E5 Where two or more test portions have been analysed, the arithmetic mean of the most accurate results obtained from each portion should be reported. Where good comminution and/or mixing of samples has been undertaken, the RSD of replicate results of the test portions should normally not exceed 30% for residues significantly above the LOQ. Close to the LOQ, the variation may be higher and additional caution is required in deciding whether or not a limit has been exceeded. Alternatively, the limits for repeatability, or reproducibility, given in Reg. 546/2011, may be applied, although these do not incorporate sub-sampling error (which is particularly important when undertaking dithiocarbamate or fumigant analyses).

E6 In general, residues data do not have to be adjusted for recovery when the mean recovery is within the range of 70-120%. If residues data are adjusted for recovery, then this must be stated. Exceedances of the MRL must be supported by individual recovery results (from the same batch) within the range of the mean recovery (70-120%)  $\pm 2 \times RSD$ , at least for the repeat confirmatory analyses. If recovery within this range cannot be achieved, enforcement action is not necessarily precluded, but the risk of relatively poor accuracy must be taken into account. It is then highly recommended to correct for recovery, preferably by using standard addition or isotopically labeled standards, for all cases of MRL exceedances.

## Rounding of data

E7 It is essential to maintain uniformity in reporting results of residue levels. In general, results above the RL and <10 mg/kg should be rounded to two significant figures. Results  $\geq$ 10 mg/kg may be rounded to three significant figures or to a whole number. Reporting limits should be rounded to 1 significant figure at <10 mg/kg and two significant figures at  $\geq$ 10 mg/kg. These rounding rules do not necessarily reflect the uncertainty associated with the reported data. Additional significant figures may be recorded for the purpose of statistical analysis and when reporting results for proficiency tests. In some cases the rounding may be specified by, or agreed with the customer/stakeholder of the control or monitoring program. In any case, the rounding of results should never lead to a different decision being taken with regard to the exceedance of a legal limit such as the MRL. Thus, rounding to significant figures should be done after the final calculation of the result.

## Qualifying results with measurement uncertainty

E8 It is a requirement under ISO/IEC 17025 that laboratories determine and make available the (expanded) measurement uncertainty (MU), expressed as U', associated with analytical results. Laboratories should have sufficient repeatability/reproducibility data from method validation/verification, inter-laboratory studies (e.g. proficiency tests), and in-house quality control tests, which can be used to estimate the MU<sup>1</sup>.

The MU describes the range around a reported or experimental result within which the true value can be expected to lie within a defined probability (confidence level). MU ranges must take into consideration all sources of error.

MU data<sup>2</sup> should be applied cautiously to avoid creating a false sense of certainty E9 about the true value. Estimates of typical MU are based on previous data and may not reflect the MU associated with the analysis of a current sample. Typical MU may be estimated using an ISO (Anonymous 1995,'Guide to the expression of uncertainty in measurement' ISBN 92-67-10188-9) or Eurachem<sup>3</sup> approach. Reproducibility RSD (or repeatability RSD if reproducibility data are not available) may be used, but the contribution of additional uncertainty sources (e.g. heterogeneity of the laboratory sample from which the analytical test portion has been withdrawn) due to differences in the procedures used for sample preparation, sample processing and sub-sampling should be included. Extraction efficiency and differences in standard concentrations should also be taken into account. MU data relate primarily to the analyte and matrix used and should be extrapolated to other analyte/matrix combinations with caution. MU tends to increase at lower residue levels, especially as the LOQ of a method is approached. It may therefore be necessary to generate MU data over a range of residue levels to reflect those typically found during routine analysis.

E10 Another practical approach for a laboratory to verify its MU estimation, based on its own within-laboratory data, is by evaluating its performance in recent proficiency tests (see Appendix C). Proficiency test results can provide an important indication of the contribution of the inter-laboratory bias to the MU of an individual laboratory. Replicate analyses of a specific sample, combined with concurrent recovery determinations, can improve the accuracy of an individual-laboratory result and improve the estimate of MU. These uncertainty data will include the repeatability of sub-sampling and analysis, but not interlaboratory bias. This practice will be typically applied when the analytical results are extremely important (e.g. an MRL compliance check).

<sup>&</sup>lt;sup>1</sup> Codex Alimentarius Commission Guideline CAC/GL 59-2006, Guidelines on estimation of uncertainty of results.

<sup>&</sup>lt;sup>2</sup> L. Alder et al., Estimation of measurement uncertainty in pesticide residue analysis. J. AOAC Intern., 84 (2001) 1569-1577.

<sup>&</sup>lt;sup>3</sup> EURACHEM/CITAC Guide, Quantifying uncertainty in analytical measurement, 3rd Edition, 2012, http://www.eurachem.org/images/stories/guides/pdf/QUAM2012\_P1.pdf

E11 The use of reporting limits based on the lowest validated spike level during method validation eliminates the need to consider uncertainty associated with residue levels found <RL.

## Interpretation of results for enforcement purposes

E12 Assessment of whether or not a sample contains a residue which is an MRL exceedance is generally only a problem in cases where the level is relatively close to the MRL. The decision should take account of concurrent AQC data and the results obtained from replicate test portions, together with any assessment of typical MU. The possibility of residue loss or cross-contamination having occurred before, during, or after sampling, must also be considered.

E13 A default expanded MU of 50% (corresponding to a 95% confidence level and a coverage factor of 2) has been calculated from EU proficiency tests. In general the 50% value covers the inter-laboratory variability between the European laboratories and is recommended to be used by regulatory authorities in cases of enforcement decisions (MRL-exceedances). A prerequisite for the use of the 50% default expanded MU is that the laboratory must demonstrate that its own expanded MU is less than 50%. In cases where an exceedance of an MRL is also an exceedance of the acute reference dose, an expanded MU with a lower confidence level can be applied as a precautionary measure.

E14 If laboratories experience individual cases of unacceptably high repeatability, or within-laboratory reproducibility-RSDwR (e.g. at very low concentration levels), or unsatisfactory z-scores during proficiency tests, the use of a correspondingly higher MU figure must be considered. For results obtained with single-residue methods, particularly if stable isotopically labelled internal standards are used, lower expanded MU can be justified, especially if supported by correspondingly better between-laboratory reproducibility RSDR (< 25%).

E15 If required, the result should be reported together with the expanded MU as follows: Result =  $x \pm U$  (units), with x representing the measured value. For official food control by regulatory authorities, compliance with the MRL must be checked by assuming that the MRL is exceeded if the measured value exceeds the MRL by more than the expanded uncertainty (x – U > MRL). With this decision rule, the value of the measurand is above the MRL with at least 97.5% confidence<sup>4</sup>. Thus, the MRL is exceeded if x-U > MRL. E.g., in case the MRL = 1 and x = 2.2, then x-U = 2.2 – 1.1 (= 50% of 2.2), which is > MRL.

<sup>&</sup>lt;sup>4</sup> Eurachem/CTAT Guide, Use of uncertainty information in compliance assessment, 1<sup>st</sup> Edition, 2007.

## F. Pesticide standards, stock solutions and calibration solutions

Identity, purity, and storage of "pure" standards

F1 "Pure" reference standards of analytes should be of known purity and each must be uniquely identified and the date of receipt recorded. They should be stored at low temperature, preferably in a freezer, with light and moisture excluded, i.e. under conditions that minimise the rate of degradation. Under such conditions, the supplier's expiry date, which is often based on less stringent storage conditions, may be replaced, as appropriate for each standard, by a date allowing for storage up to 10 years. The "pure" standard may be retained and a new expiry date allocated, providing that it is checked by the appropriate date and its purity is shown to remain acceptable. Ideally, the identity of a freshly acquired "pure" standard should be checked if the analytes new to the laboratory. For screening purposes only, the "pure" standards and derived solutions may be used after the expiry date, providing that the reporting level can be achieved. If the pesticide has been detected, a new or certified "pure" standard and calibration solution made thereof has to be used for quantification.

## Preparation and storage of stock standards

F2 When preparing stock standards (solutions, dispersions or gaseous dilutions) of "pure" standards (analytes and internal standards) the identity and mass (or volume, for highly volatile compounds) and the identity and amount of the solvent (or other diluents) must be recorded. The solvent(s) must be appropriate to the analyte (solubility, no chemical reactions) and method of analysis. Moisture must be excluded during equilibration of the "pure" standard to room temperature before use and concentrations must be corrected for the purity of the "pure" standard.

F3 Not less than 10 mg of the "pure" standard should be weighed using a 5 decimal place balance. The ambient temperature should be that, at which the glassware is calibrated, otherwise preparation of the standard should be based on mass measurement. Volatile liquid analytes should be dispensed by volume or weight (if the density is known) directly into solvent. Gaseous (fumigant) analytes may be dispensed by bubbling into solvent and weighing the mass transferred, or by preparing gaseous dilutions (e.g. with a gas-tight syringe, avoiding contact with reactive metals).

F4 Stock standards must be labelled indelibly, allocated an expiry date and stored at low temperature in the dark in containers that prevent any loss of solvent and entry of water. Following equilibration to room temperature, solutions must be re-mixed and a check made to ensure that the analyte remains completely dissolved, especially where solubility at low temperatures is limited. The use of a different solvent, different storage conditions or the preparation of stock solutions with lower concentration can help to overcome this problem. The stability of pesticides may depend on the solvent used. Currently available data show that stock standards solutions of the large majority of pesticides, when stored in tightly closed glass containers in the freezer, are sufficiently stable for at least 5 years in toluene or acetone and for at least 3 years in acetonitrile, methanol or ethyl acetate.

F5 For suspensions (e.g. dithiocarbamates) and solutions (or gaseous dilutions) of highly volatile fumigants that should be prepared freshly, the accuracy of the solution should be compared with a second solution made independently at the same time.

## Preparation, use and storage of working standards

F6 When preparing working standards, a record must be kept of the identity and amount of all solutions and solvents employed. The solvent(s) must be appropriate to the analyte (solubility, no chemical reactions) and method of analysis. The standards must be labelled indelibly, allocated an expiry date and stored at low temperature in the dark in containers that prevent any loss of solvent and entry of water. Septum closures are particularly prone to evaporation losses (in addition to being a potential source of contamination) and should be replaced as soon as practicable after piercing, if solutions are to be retained. Following equilibration to room temperature, solutions must be re-mixed and a check made to ensure that the analyte remains in solution, especially where solubility at low temperatures is limited.

F7 At method development or validation, or for analytes new to the laboratory, the response detected should be shown to be due to the analyte, rather than to an impurity or artefact. If degradation of the analyte occurs during extraction, clean-up or separation, and the degradation product is commonly found in samples, but is excluded from the residue definition, then the results must be confirmed using alternative techniques that avoid this problem.

## Testing and replacement of standards

F8 The stability of an existing and possibly expired "pure" standard may be checked by preparing a new stock standard and comparing the detector responses. The comparison should be undertaken using appropriate dilutions of individual standards or mixtures of standards. Inexplicable differences in apparent concentration between old and new standards must be investigated.

F9 The means from at least three replicate measurements for each of two solutions (old and new) should not normally differ by more than  $\pm 10\%$ . The mean value from the new solution is taken to be 100% and is also used as a basis for the calculation of the percentage-difference. If the mean response of the old standard differs by more than  $\pm 10\%$  from the new standard, then storage time or conditions may have to be adjusted. Both old and new solution should be checked against another new solution that is prepared independently from the first two.

F10 The variability of (preferably at least 5) replicate injections (expressed as repeatability-RSDr) should also be taken into account and typically not exceed 10% for both the old and new standard solution. Where it is known that an analyte is difficult to analyse (e.g. because of degradation), a maximum RSDr of 15% may still be accepted.

F11 Discrepancies between the concentrations of new and old solution can be due to a number of factors other than analyte degradation. Analyte precipitation, solvent evaporation, differences in the purities between the old and new standards, errors in weighing, or errors in the instrumental analysis. The use of an internal standard may help to reduce measurement variation and thus the number of replicate injections required (minimum 3) to comply with the  $\pm 10\%$  or  $\pm 15\%$  differences. Where sufficient evidence exists (data from  $\geq 2$  other labs) that a certain pesticide is stable using specified storage conditions (time, solvent, temperature etc.) then other laboratories reproducing these storage conditions can reduce their own stability checks. However, possible solvent evaporation must be checked gravimetrically. In some cases certain additives (e.g. acids) may have to be added to stock solutions to prevent degradation of the analytes.

## G. Analytical method validation and performance criteria

## Quantitative methods

G1 Within-laboratory method validation should be performed to provide evidence that a method is fit for the purpose for which it is to be used. Method validation is a requirement of accreditation bodies, and must be supported and extended by method performance verification during routine analysis (analytical quality control and on-going method validation). Where practicable, all procedures (steps) that are undertaken in a method should be validated.

G2 Representative matrices may be used to validate multi- and single- residue methods. As a minimum, one representative commodity from each commodity group as described in Annex A must be validated, depending on the intended scope of the method. When the method is applied to a wider variety of matrices, complementary validation data should be acquired, e.g. from on-going QC during routine analyses. An example of a practical approach to the validation procedure is presented in Appendix A.

G3 The method must be tested to assess sensitivity, mean recovery (as a measure of trueness or bias), precision (as repeatability RSDr) and method-LOQ. A minimum of 5 replicates is required (to check the recovery and precision) at the targeted LOQ or reporting limit of the method, and at least one other higher level, for example, 2-10x the targeted LOQ or the MRL. Where the residue definition includes two or more analytes, then wherever possible, the method should be validated for all analytes included in the residue definition.

G4 If the analytical method does not permit determination of recovery (for example, direct analysis of liquid samples, SPME, or headspace analysis), the precision is determined from repeat analyses of calibration standards. The bias is usually assumed to be zero, although this is not necessarily the case. In SPME and headspace analysis, the trueness and precision of calibration may depend on the extent to which the analyte has equilibrated with respect to the sample matrix. Where methods depend upon equilibrium, this must be demonstrated during method development.

## Method performance acceptability criteria

G5 A quantitative analytical method should be demonstrated at both initial and extended validation stages, as being capable of providing acceptable mean recovery values at each spiking level and for at least one representative commodity from each relevant group (see Annex 1). Acceptable mean recoveries are those within the range 70–120%, with an associated repeatability RSDr  $\leq$  20%, for all compounds within the scope of a method. The method-LOQ is the lowest spike level of the validation meeting these method performance acceptability criteria. In certain cases and typically with multi-residue methods, recoveries outside this range may be accepted. Exceptionally, where recovery is low but consistent (i.e. demonstrating good precision) and the basis for this is well established (e.g. due to analyte distribution in a partitioning step), a mean recovery below 70% may be acceptable. However, a more accurate method should be used, if practicable. Within-laboratory reproducibility (RSD<sub>wR</sub>), which may be determined from on-going QC-data in routine analyses, should be  $\leq$  20%, excluding any contribution due to sample heterogeneity.

## Screening methods

G6 Screening methods, especially those involving automated MS-based detection, offer laboratories a cost-effective means to extend their analytical scope to analytes which potentially have a low probability of being present in the samples. Analytes that occur more frequently should continue to be sought and measured using validated quantitative multiresidue methods. G7 For screening methods the confidence of detection of an analyte at a certain concentration level should be established. This can be achieved using screening methods based on the RL from the validation of a quantitative method or screening methods based on the screening detection limit (SDL) from the validation of a qualitative method.

G8 When using a screening method, the calibration standard solution corresponding to the RL or SDL should be placed, at least at the beginning and the end of the sample sequence to ensure that the analytes remain detectable throughout the whole batch of samples in the sequence. When a pesticide is detected, it can only be tentatively reported. A subsequent confirmatory analysis using a validated quantitative method, including an appropriate calibration procedure, must be applied before a reliable qualitative result may be reported. If an analyte is not detected, then the result must be reported as <SDL mg/kg or <RL mg/kg.

G9 The validation of a screening method based on a SDL can be focused on detectability. For each commodity group (see Annex 1), a basic validation should involve analysis of at least 20 samples spiked at the estimated SDL. The samples selected should represent multiple commodity categories from the commodity group, with a minimum of two different samples for each commodity category and should be representative for the intended scope of the laboratory. Additional validation data can be collected from on-going AQC-data and method performance verification during routine analysis.

## Method performance acceptability criteria

G10 When the screening method is only intended to be used as a qualitative method, there are no requirements with regard to recovery of the analytes. In order to determine the selectivity, the presence of false detects should be verified using non-spiked (preferably "blank") samples. Provided the analytes that are tentatively detected by the screening method are identified and confirmed by a second analysis of the sample using an appropriate confirmatory method, there is no need for a strict criterion for the number of false detects. The SDL of the qualitative screening method is the lowest level at which an analyte has been detected (not necessarily meeting the MS-identification criteria) in at least 95% of the samples (i.e. an acceptable false-negative rate of 5%).

G11 For analytes that have not been included in the initial or on-going method validation, the confidence level of detection at a certain residue level will not be known. Consequently analytes outside of the scope of validation can be detected using the method, but no SDL can be specified.

G12 When using a qualitative screening method, only analytes that have been validated can be added to the routine scope of the laboratory.

## H. Additional recommendations

## Contamination

H1 Samples must be separated from each other and from other sources of potential contamination, during transit to, and storage at the laboratory. This is particularly important with surface or dusty residues, or with volatile analytes. Samples known, or thought, to bear such residues should be doubly sealed in polythene or nylon bags and transported and processed separately.

H2 Volumetric equipment, such as flasks, pipettes and syringes must be cleaned scrupulously, especially before re-use. As far as practicable, separate glassware, etc., should be allocated to standards and sample extracts, in order to avoid cross-contamination. The use of excessively scratched or etched glassware should be avoided. Solvents used for fumigant residues analysis should be checked to ensure that they do not contain target analyte(s).

H3 Where an internal standard is used, unintended contamination of extracts or analyte solutions with the internal standard, or vice versa, must be avoided.

H4 Where the analyte occurs naturally, or as a contaminant, or is produced during the analysis (e.g. biphenyl in herbs, inorganic bromide in all commodities, sulphur in soil, or carbon disulfide produced from the Brassicaceae), low-level residues from pesticide use cannot be distinguished from background levels. Natural occurrence of these analytes must be considered in the interpretation of results. Dithiocarbamates, ethylenethiourea or diphenylamine can occur in certain types of rubber articles and this source of contamination must be avoided.

## Interference

H5 Equipment, containers, solvents (including water), reagents, filter aids, etc., should be checked as sources of possible interference. Rubber and plastic items (e.g. seals, protective gloves, and wash bottles), polishes and lubricants are frequent sources of interferences. Vial seals should be PTFE-lined. Extracts should be kept out of contact with seals, especially after piercing, for example, by keeping vials upright. Vial seals may have to be replaced quickly after piercing, if re-analysis of the extracts is necessary. Analysis of reagent blanks should identify sources of interference in the equipment or materials used.

H6 Matrix effects or matrix interferences from natural constituents of samples are frequent. The interference may be peculiar to the determination system used, variable in occurrence and intensity, and may be subtle in nature. If the interference takes the form of a response overlapping that of the analyte, a different clean-up or determination system may be required. Matrix effects in terms of suppression or enhancement of the detection system response is dealt with in paragraph C22. If it is not practicable to eliminate matrix effects or to compensate for such effects by matrix-matched calibration, the overall accuracy of analysis should nonetheless comply with the criteria in paragraphs C45 and E6.

## Annex A Commodity groups and representative commodities<sup>5</sup>

Commodity groups	Typical commodity categories	Typical representative commodities			
1. High water content	Pome fruit	Apples, pears			
	Stone fruit	Apricots, cherries, peaches,			
	Other fruit	Bananas			
	Alliums	Onions, leeks			
	Fruiting	Tomatoes, peppers, cucumber, melon			
	vegetables/cucurbits				
	Brassica vegetables	Cauliflower, Brussels-sprouts, cabbage, broccoli			
	Leafy vegetables and fresh herbs	Lettuce, spinach, basil			
	Stem and stalk vegetables	Celery, asparagus			
	Forage/fodder crops	Fresh alfalfa, fodder vetch, fresh sugar beets			
	Fresh legume vegetables	Fresh peas with pods, peas, mange tout, broad beans, runner beans, French beans			
	Leaves of root and tuber vegetables	Sugar beet and fodder beet tops			
	Fresh Fungi	Champignons, canterelles			
	Root and tuber	Sugar beet and fodder beet roots, carrots,			
	vegetables or feed	potatoes, sweet potatoes			
2. High acid content	Citrus fruit	Lemons, mandarins, tangerines, oranges			
and high water content <sup>(1)</sup>	Small fruit and berries	Strawberry, blueberry, raspberry, black currant, red currant, white currant, grapes			
	Other	Kiwifruit, pineapple, rhubarb			
3. High sugar and low water content <sup>(2)</sup>	Honey, dried fruit	Honey, raisins, dried apricots, dried plums, fruit jams			
4a. High oil content and	Tree nuts	Walnuts, hazelnuts			
very low water content	Oil seeds	Oilseed rape, sunflower, cotton-seed, soybeans, peanuts, sesame etc.			
	Pastes of tree nuts and oil seeds	Peanut butter, tahina, hazelnut paste			
	Oils from tree nuts, oil seeds and oily fruits	Olive oil, rapeseed oil, sunflower oil, pumpkin seed oil			
4b. High oil content and intermediate water content	Oily fruits and products	Olives, avocados and pastes thereof			
5. High starch and/or protein content and low water and fat	Dry legume vegetables/pulses	Field bean, dried broad bean, dried haricot bean (yellow, white/navy, brown, speckled), lentils			
content	Cereal grain and products thereof	Wheat, rye, barley and oat grain; maize, rice Wholemeal bread, white bread, crackers, breakfast cereals, pasta			
6. "Difficult or unique commodities"		Hops Cocoa beans and products thereof, coffee, tea			
7 Moat (muscle) and	Pod musele	Spices			
7. Meat (muscle) and	Red muscle	Beef, pork, lamb, game, horse			
Seafood	White muscle	Chicken, duck, turkey			
		Liver, kidney			
	Fish	Cod, haddock, salmon, trout			
	Crustaceans	Shrimp, scallop, crab			

Vegetable and fruits, cereals and food of animal origin

<sup>&</sup>lt;sup>5</sup> On the basis of OECD Environment, Health and safety Publications, Series on Testing and Assessment, No72 and Series of Pesticides No39

Commodity groups	Typical commodity categories	Typical representative commodities			
8. Milk and milk	Milk	Cow, goat and buffalo milk			
products	Cheese	Cow and goat cheese			
	Dairy products	Yogurt, cream			
9. Eggs	Eggs	Chicken, duck, quail and goose eggs			
10. Fat from food of	Fat from meat	Kidney fat, lard			
animal origin <sup>(3)</sup>	Milk fat <sup>(4)</sup>	Butter			
	Fish oil	Cod liver oil			

<sup>(1)</sup> If a buffer is used to stabilise the pH changes in the extraction step, then commodity Group 2 can be merged with commodity Group 1.

(2) Where commodities of Group 3 are mixed with water prior to extraction to achieve a water content of >70%, this commodity group may be merged with Group 1. The RLs should be adjusted to account for smaller sample portions (e.g. if 10g portions are used for commodities of Group 1 and 5g for Group 3, the RL of Group 3 should be twice the RL of Group 1 unless a commodity belonging to Group 3 is successfully validated at a lower level).

<sup>13</sup> "Difficult commodities" should only be fully validated if they are frequently analysed. If they are only analysed occasionally, validation may be reduced to just checking the reporting limits using spiked blank extracts.

<sup>(4)</sup> If methods to determine non-polar pesticides in commodities of Group 7 are based on extracted fat, these commodities can be merged with Group 10.

## Appendix A. Method validation procedure: outline and example approaches

Validation is undertaken following the completion of the method development or before a method that has not been previously used is to be introduced for routine analysis. We distinguish between initial validation of a quantitative analysis method to be applied in the laboratory for the first time and to extension of the scope of an existing validated method for new analytes and matrices.

## Quantitative analysis

## 1. Initial full validation

Validation needs to be performed

- for all analytes within the scope of the method
- for at least 1 commodity from each of the commodity groups (as far as they are within the claimed scope of the method or as far as applicable to samples analysed in the laboratory)

## Experimental:

A typical example of the experimental set up of a validation is:

<u>Sample set (sub-samples from 1 homogenised sample):</u>

Reagent blank 1 blank (non-spiked) sample 5 spiked samples at target LOQ 5 spiked samples at 2 10x target LOQ

5 spiked samples at 2-10x target LOQ

Instrumental sample sequence: Calibration standards in solvent Calibration standards in matrix Reagent blank Blank sample 5 spiked samples at target LOQ 5 spiked samples at 2-10 x target LOQ Calibration standards in matrix

Spiking of commodities is a critical point in validation procedures. In general the spiking procedure should reflect as much as possible the circumstances during routine application of the method. If for example, samples are milled cryogenically and extracted in frozen condition spiking should be done on frozen portions of blank material and extracted immediately. If samples are milled at room temperature and extracted on average after 20 min, spiking should be done on blank homogenates at room temperature and extracted after 20 minutes waiting time. In general, spiking of samples will not simulate incurred residues even if the spiked sample is left standing for a certain time. To study the extractability of incurred residues real samples should be taken.

## Data evaluation:

Inject the sample sequence, calibrate and and quantify as is described in this AQC document.

Evaluate the parameters from Table 1 and verify them against the criteria.

## Table 1. Validation parameters and criteria.

Parameter What/how		Criterion	Cross reference to AQC document	
Linearity	Calibration curve	Residuals < ±20%	C14-C18	
Matrix effect	Comparison of response from solvent standards and matrix-matched standards	Assess % matrix effect	C22-C24	
LOQ	Lowest spike level meeting the method performance criteria for trueness and precision	≤MRL	G2-G5	
Specificity	Response in reagent blank and blank control samples		H5	
Trueness (bias)	Average recovery for spike levels tested	70-120%	C45	
Precision (RSD <sub>r</sub> )	Repeatability RSD <sub>r</sub> for spike levels tested	≤ 20%	E14,G6	
Precision (RSD <sub>wR</sub> )	Within-laboratory reproducibility, derived from on-going method validation / verification	≤ 20%		
Robustness	Average recovery and RSD <sub>wR</sub> , derived from on-going method validation / verification	See above	G2, G4	

## 2. Extension of the scope of the method: new analytes

New analytes that are added to a previously validated method need to be validated using the same procedure as outlined above for initial validation.

Alternatively, the validation of new analytes can be integrated in the on-going quality control procedure. As an example: with each batch of routine samples one or more commodities from the applicable commodity category are fortified at LOQ and one higher level. Determine recovery and occurrence of any interference in the corresponding unfortified sample. When for both levels 5 recovery values have been collected, the average recovery and within -laboratory reproducibility (RSD<sub>wR</sub>) can be determined and tested against the criteria in Table 1.

## 3. Extension of the scope of the method: new matrices

A pragmatic way of validation of the applicability of the method to other matrices from the same commodity group is to do this during the on-going quality control performed concurrently with analysis of the samples. See below.

## 4. On going performance validation / verification

The purpose of on-going method validation is to:

- demonstrate robustness through evaluation of mean recovery and within -laboratory reproducibility (RSD\_{wR})
- demonstrate that minor adjustments made to the method over time do not unacceptably affect method performance
- demonstrate applicability to other commodities from the same commodity category (see also above)
- determine acceptable limits for individual recovery results during routine analysis
- collect information for estimation of the within-laboratory measurement uncertainty

## Experimental:

Typically, with each batch of samples routinely analysed, one or more samples of different commodities from the applicable commodity category are fortified with the analytes and analysed concurrently with the samples.

#### Data evaluation:

Determine for each analyte the recovery from the fortified sample and occurrence of any interference in the corresponding unfortified sample. Periodically (e.g. annually) determine average recovery and reproducibility (RSDwR) and verify data obtained against the criteria from Table 1. These data can also be used to set or update limits for acceptability of individual recovery determinations as outlined in paragraph of the AQC document and for estimation of the measurement uncertainty.

## Appendix B. Examples of conversion factors.

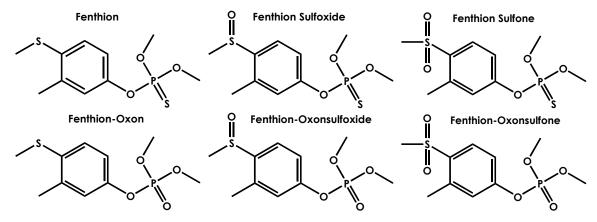
The MRL residue definitions for a number of pesticides include not only the parent pesticide, but also its metabolites or other transformation products.

In Example 1, the sum of the components is expressed as fenthion, following adjustment for the different molecular weights (conversion factors), in Example 2, the sum of triadimenon and triadimenol is expressed as their arithmetic sum, and in Example 3, the sum of thiodicarb and methomyl is expressed as methomyl.

The following examples illustrate the three different types of summing that are required in order to meet the requirements of the residue definition.

## Example 1.

Fenthion, its sulfoxide and sulfone, and their oxygen analogues (oxons), all appear in the residue definition and all should be included in the analysis.



Example of calculating the conversion factor (Cf)

CFenthionSO to Fenthion = MWFenthion/MWFenthionSO x CFenthion SO = 278.3/294.3 x CFenthion SO = 0.946 x CFenthionSO

Compound			Mw	Cf
Fenthion	RR´S	P=S	278,3	1,00
Fenthion sulfoxide	rr'so	P=S	294,3	0,946
Fenthion sulfone	RR´SO2	P=S	310,3	0,897
Fenthion oxon Fenthion oxon sulfoxide Fenthion oxon sulfone	RR´S RR´SO R´SO2	P=0 P=0 P=0	262,3 278,3 294,3	1,06 1,00 0,946

<u>Residue Definition:</u> Fenthion (fenthion and its oxygen analogue, their sulfoxides and sulfones expressed as fenthion)

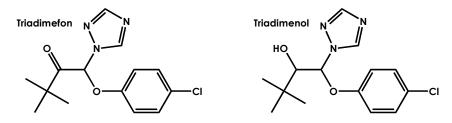
Where the residue is defined as the sum of the parent and transformation products, the concentrations of the transformation products should be adjusted according to their molecular weight being added to the total residue concentration.

CFenthionSum = 1.00 x CFenthion + 0.946 x CFenthion SO + 0.897 xCFenthion SO2 +

1.06 x CFenthionoxon + 1.00x CFenthionoxon SO + 0.946 x CFenthionoxon SO2

## Example 2.

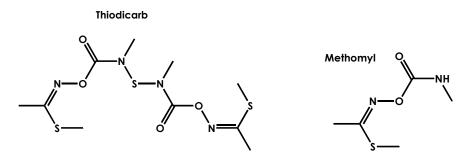
Residue Definition: Triadimefon and triadimenol (sum of triadimefon and triadimenol)



C Triadimeton and triadimenolSum =  $1.00 \times C$  Triadimeton +  $1.00 \times C$  Triadimenol

## Example 3

<u>Residue Definition:</u> Methomyl and Thiodicarb (sum of methomyl and thiodicarb, expressed as methomyl)



CMethomylSum = CMethomyl + C Thiodicarb X (2XMWMethomyl / MWThiodicarb) =

= (2x162.2 / 354.5) x C<sub>Thiodicarb</sub> = 0.915 x C <sub>Thiodicarb</sub>

C Methomyl Sum = C Methomyl +0.915 \* CThiodicarb

## Appendix C. Examples for the estimation of measurement uncertainty of results

In order to estimate Measurement Uncertainty (MU) of results for the determination of pesticide residues, several documents are recommended to be read that help to provide a better understanding of this topic, such as Eurachem<sup>(1)</sup>, Nordtest<sup>(2)</sup>, Eurolab<sup>(3)</sup> and Codex CAC/GL 59-2006<sup>(4)</sup> Guidelines.

Nevertheless, it has been considered useful to include an appendix with clear examples in this document <sup>(5)</sup>. Two approaches are explained in depth. In both examples, an expanded coverage factor of k = 2 is assumed to calculate the expanded MU represented by U' from the relative standard uncertainty u' in equation 1.

$$U' = k \times U$$

Equation 1

## 1st Approach:

Whenever a laboratory has participated in a number of Proficiency Tests (EUPTs or other relevant PTs on pesticide residues) and achieved acceptable z-scores for all (or almost all) the pesticides present in the test material, this approach can be applied.

In this approach, a default value of 50% as expanded MU is applied. This default value is based on the mean relative standard deviations of results reported by the participanting laboratories in a number of EUPTs for multi-residue methods on fruit and vegetables. This mean ranged around 25%, providing an expanded uncertainty of 50%.

$$U' = 2 \times 0.25 = 0.50$$

U' = 50%

The first approach is to be adopted, providing that the MU of the laboratory is  $\leq$  50% and in order to do this the 2<sup>nd</sup> approach can be undertaken.

## 2<sup>nd</sup> Approach:

In this approach, the expanded MU is calculated using the within-laboratory reproducibility relative standard deviation combined with estimates of the method and the laboratory bias using PT data<sup>(2)</sup> applying equation 2.

$$\upsilon' = \sqrt{\upsilon' (RSD_{wR})^2 + \upsilon' (bias)^2}$$

Equation 2

In equation 2:

υ´	is the combined standard uncertainty			
u´(RSD <sub>wr</sub> )	is the within-laboratory reproducibility			
u´(bias)	is the uncertainty component arising from method and laboratory			
	bias, estimated from PT data.			

To calculate  $u'(RSD_{wR})$  preferably long-term quality control (QC) recovery data should be used although recoveries coming from validation data can be included too.

Note: within-laboratory variability coming from calibration is considered to be included in the long-term quality control recovery variability<sup>(1)</sup>.

The standard deviation of all the recoveries percentage taken into account is calculated.

For the example presented here, validation recoveries are taken for all pesticides that have been validated in the same multi residue method (MRM) and for which the laboratory is used to take part in the PTs. Also the long-term QC recovery data in the range of 60%-140% are included for two different levels and for the fruit and vegetables matrices normally analysed in the laboratory. A minimum of 31 results must be taken into  $account^{(4)}$ . For two methods: one for LC with 93 pesticides and the other for GC with 66 pesticides, the standard deviation of all the recovery percentages is 0.15. The  $u'(RSD_{wR})$  is therefore 0.15.

The u'(bias) component is calculated from the performance of the laboratory in PT studies as stated in many guidelines<sup>(2-4)</sup>. Participation of EU official laboratories in the EUPTs is mandatory. therefore taking results from at least 2 EUPT-FV will provide enough data (above 31 results) to conduct this approach.

For this example, the 2 EUPT-FV results reported are in total 39 pesticide results. From these two PTs the information that needs to be used is the assigned value or median, the real dispersion of results reported by the laboratories for each of the pesticides present in the sample (the Qn or robust standard deviation) and the number of laboratories reporting quantitative results for those pesticides.

Table 1 shows the number of the EUPT-FV wherein the lab has participated (column A), the pesticides reported (column B), the pesticide concentration reported (column C), the assigned value or median (column D), the square of the bias (column E) which is [(column C – column D) / (column D)]<sup>2</sup>, then the dispersion of the data from the participants or Qn (column F), then the number of laboratories reporting results for each of the pesticides (column G), then the square root of column G (column H) and then the coefficient between column F and column H (column I).

Α	В	с	D	E	F	G	н	I
EUPT-FV	Pesticides	Lab Results	PT Assigned Values	(bias´i)²	Qn	No. Results	√No.	$\frac{Qn}{\sqrt{No.}}$
	Acetamiprid	0.337	0.419	0.0383	0.18	85	9.220	0.020
	Boscalid	0.139	0.238	0.1720	0.22	74	8.602	0.026
	Chlorpyrifos-methyl	0.056	0.078	0.0796	0.26	126	11.225	0.023
	Diazinon	0.412	0.603	0.1003	0.24	125	11.180	0.021
	Endosulfan Sulphate	0.062	0.102	0.1538	0.29	110	10.488	0.028
	Hexythiazox	0.396	0.509	0.0493	0.29	80	8.944	0.032
	Isofenphos-methyl	0.436	0.499	0.0159	0.17	69	8.307	0.020
	Kresoxim-methyl	0.028	0.050	0.1936	0.22	113	10.630	0.021
EUPT-FV-10	Malathion	0.697	0.771	0.0091	0.32	124	11.136	0.029
Carrot	Methamidophos	0.245	0.342	0.0798	0.37	103	10.149	0.036
	Methiocarb	0.096	0.157	0.1510	0.31	65	8.062	0.038
	Methomyl	0.538	0.739	0.0740	0.22	88	9.381	0.023
	Oxamyl	0.274	0.322	0.0222	0.19	84	9.165	0.021
	Pendimethalin	0.056	0.074	0.0592	0.21	96	9.798	0.021
	Phosmet	0.139	0.236	0.1689	0.28	95	9.747	0.029
	Quinoxyfen	0.244	0.298	0.0328	0.23	95	9.747	0.024
	Triadimenol	0.265	0.331	0.0398	0.27	103	10.149	0.027
	Vinclozolin	0.90	1.04	0.0181	0.24	124	11.136	0.022
	Aldicarb	0.679	0.658	0.0010	0.20	91	9.539	0.021
	Azinphos-methyl	0.349	0.355	0.0003	0.28	128	11.314	0.025
	Boscalid	0.373	0.414	0.0098	0.25	102	10.100	0.025
	Buprofezin	0.453	0.638	0.0841	0.30	118	10.863	0.028
	Cadusafos	0.810	0.611	0.1061	0.24	76	8.718	0.028
	Carbofuran	0.245	0.283	0.0180	0.20	107	10.344	0.019
	Deltamethrin	0.138	0.157	0.0146	0.25	130	11.402	0.022
	Diazinon	1.140	1.25	0.0077	0.26	144	12.000	0.022
	Isofenphos-methyl	0.498	0.54	0.0060	0.24	86	9.274	0.026
	Lambda-cyhalothrin	0.211	0.266	0.0428	0.24	138	11.747	0.020
EUPT-FV-11 Cauliflower	Metalaxyl	0.445	0.45	0.0001	0.21	122	11.045	0.019
Caulinower	Methamidophos	0.341	0.4045	0.0246	0.33	109	10.440	0.032
	Methidathion	0.453	0.472	0.0016	0.24	136	11.662	0.021
	Methomyl	0.190	0.277	0.0986	0.18	84	9.165	0.020
	Monocrotophos	0.322	0.4375	0.0697	0.21	95	9.747	0.022
	Oxamyl	0.230	0.2485	0.0055	0.17	89	9.434	0.018
	Parathion-methyl	0.277	0.32	0.0181	0.24	129	11.358	0.021
	Phosalone	0.383	0.368	0.0017	0.30	136	11.662	0.026
	Procymidone	0.750	0.78	0.0015	0.20	136	11.662	0.017
	Thiacloprid	0.961	0.879	0.0087	0.15	82	9.055	0.017
	Triazophos	0.612	0.538	0.0189	0.30	132	11.489	0.026
$\sum (bias)^2$				1.09973		Σ	$\frac{Qn}{\sqrt{No.}}$	0.9326
No. of Results (m) 3				esults (m) 39	No. of Results (m) 39			

Then equation 3 is used:

$$U' = \sqrt{RMS_{bias}^{2} + U'(C_{ref})^{2}}$$
 Equation 3

Where:

• RMS' bias is the Root Mean Square of the sum of the squared bias [(sum of column E) divided by the number of results taken from the PTs (m =39)] as indicated in equation 4.

$$RMS_{bias} = \sqrt{\sum \frac{(bias)^2}{m}} = \sqrt{\frac{1.09973}{39}} = 0.2263$$
 Equation 4

u'(Cref) is an estimation of an average over several PTs. It is calculated as the sum of the Qn divided by the square root of the number of results reported by the laboratories for each of the pesticides in the scope (column I), then divided by the number of results (m) taken from the PTs (39) and multiplied by a factor of 1.253 according to ISO 13528<sup>(6)</sup>. This ISO states that u'(Cref) must be multiplied by this factor, whenever the assigned value in PTs is the median. Is calculated following equation 5.

$$U'(c_{ref}) = \frac{\sum_{i} \frac{Qn}{\sqrt{No.}}}{m} \times 1.253 = \frac{0.9326}{39} \times 1.253 = 0.0239$$
 Equation 5

When entering the results from equation 4 and 5 into iequation 3, we get the u'(bias):

$$U'(bias) = \sqrt{RMS'_{bias}^2 + U'(c_{ref})^2} = \sqrt{0.2263^2 + 0.0239^2} = 0.2283$$

Note: the u'(bias) can be calculated from the participation of the laboratory in other PTs.

Now, back to equation 2 and entering the  $u'(RSD_{wR}) = 0.15$  and the u'(bias):

$$U' = \sqrt{U'(RSD_{wR})^2 + U'(bias)^2} = \sqrt{0.15^2 + 0.2283^2} = 0.2732$$

So back to equation 1, u' = 0.27 and the expanded measurement uncertainty is therefore:

$$J' = k \times U' = 2 \times 0.27 = 0.54$$
  $U' = 54\%$ 

Both approaches have very similar results: 50% and 54%, respectively.

## References

- (1) EURACHEM Guide: Quantifying Uncertainty in Analytical Measurements, Laboratory of the Government Chemist, 2nd edition, London, 2000.
- (2) NORDTEST Report TR 537: Handbook for Calculation of Measurement Uncertainty in Environmental Laboratories, http://www.nordicinnovation.net/nordtestfiler/tec537.pdf, 2nd edition, Espoo, 2004.
- (3) EUROLAB Technical Report 1/2007: Measurement uncertainty revised: alternative approaches to uncertainty evaluation, European Federation of National Associations of Measurement, Testing and Analytical Laboratories, www.eurolab.org, Paris, 2007.
- (4) Codex Alimentarius Commission ,CAC/GL 59-2006 (Amendment 1-2011) Guidelines on Estimation of Uncertainty of Results, www.codexalimentarius.net/download/standards/10692/cxg\_059e.pdf , Rome 2006 and 2011.
- (5) P. Medina-Pastor et al., J. Agric. Food Chem., 59(2011)7609-7619
- (6) ISO 13528: Statistical methods for use in proficiency testing by interlaboratory comparisons, International Standardisation Organisation

Element Code	Element Name	Element Label	Data type⁴	Controlled terminology	Description
S.01	labSampCode	Laboratory sample code	xs:string (20)		Alphanumeric code of the analysed sample.
S.03	lang	Language	xs:string (2)	LANG	Language used to fill in the free text fields (ISO-639- 1).
S.04	sampCountry	Country of sampling	xs:string (2)	COUNTRY	Country where the sample was collected. (ISO 3166- 1-alpha-2).
S.06	origCountry	Country of origin of the product	xs:string (2)	COUNTRY	Country of origin of the product (ISO 3166-1-alpha-2 country code).
S.13	prodCode	Product code	xs:string (20)	MATRIX	Food product analysed described according to the MATRIX catalogue.
S.14	prodText	Product full text description	xs:string (250)		Free text to describe in detail the product sampled. This element becomes mandatory if "product code" is 'XXXXXA' (Not in list).
S.15	prodProdMeth	Method of production	xs:string (5)	PRODMD	Code providing additional information on the type of production for the food under analysis.
S.17	prodTreat	Product treatment	xs:string(5)	PRODTR	Used to describe the treatments or processes of the food product.
S.21	prodCom	Product comment	xs:string (250)		Additional information on the product, particularly home preparation details if available.
S.28	sampY	Year of sampling	xs:decimal (4,0)		Year of sampling.
S.29	sampM	Month of sampling	xs:decimal (2,0)		Month of sampling. If the measure is the result of a sampling over a period of time, this field should contain the month when the first sample was collected.
S.30	sampD	Day of sampling	xs:decimal (2,0)		Day of sampling. If the measure is the result of a sampling over a period of time, this field should contain the day when the first sample was collected.
S.31	progCode	Programme number	xs:string (20)		Sender's unique identification code of the programme or project for which the sample analysed was taken.
S.32	progLegalRef	Programme legal reference	xs:string (100)		Reference to the legislation for the program identified by programme number.
S.33	progSampStrategy	Sampling strategy	xs:string (5)	SAMPSTR	Sampling strategy (ref. EUROSTAT - Typology of sampling strategy, version of July 2009) performed in the programme or project identified by program code.
S.34	progType	Type of sampling program	xs:string (5)	SRCTYP	Indicate the type of programme for which the samples have been collected.
S.35	sampMethod	Sampling method	xs:string (5)	Sampmd	Code describing the sampling method
S.39	sampPoint	Sampling point	xs:string (10)	Sampnt	Point in the food chain where the sample was taken. (Doc. ESTAT/F5/ES/155 "Data dictionary of activities of the establishments").
L.01	labCode	Laboratory	xs:string (100)		Laboratory code (National laboratory code if available). This code should be unique and consistent through the transmissions.
L.02	labAccred	Laboratory accreditation	xs:string (5)	LABACC	The laboratory accreditation to ISO/IEC 17025.
R.01	resultCode	Result code	xs:string (40)		Unique identification number of an analytical result (a row of the data table) in the transmitted file. The result code must be maintained at organisation level and it will be used in further updated/deletion operation from the senders.
R.02	analysisY	Year of analysis	xs:decimal (4,0)		Year when the analysis was completed.
R.06	paramCode	Parameter code	xs:string (20)	PARAM	Parameter/analyte of the analysis described according to the Substance Code of the PARAM catalogue.
R.07	paramText	Parameter text	xs:string (250)		Free text to describe the parameter. This element becomes mandatory if "Parameter code" is ' RF- XXXX-XXX' (Not in list).

## Appendix D. List of the data elements of the Standard Sample Description

<sup>&</sup>lt;sup>6</sup>The double data type corresponds to IEEE double-precision 64-bit floating point type, the decimal represents arbitrary precision decimal numbers, the string data type represents character strings in XML. The data type xs: for double data types and other numeric data types which allow decimal separation, the decimal separator should bea "." while the decimal separator "," is not allowed.

Element Code	Element Name	Element Label	Data type <sup>6</sup>	Controlled terminology	Description
R.08	paramType	Type of parameter	xs:string (5)	PARTYP	Define if the parameter reported is an individual residue/analyte, a summed residue definition or part of a sum.
R.12	accredProc	Accreditation procedure for the analytical method	xs:string (5)	MDSTAT	Accreditation procedure for the analytical method used.
R.13	resUnit	Result unit	xs:string (5)	UNIT	All results should be reported as mg/kg.
R.14	resLOD	Result LOD	xs:double		Limit of detection reported in the unit specified by the variable "Result unit".
R.15	resLOQ	Result LOQ	xs:double		Limit of quantification reported in the unit specified by the variable "Result unit".
R.18	resVal	Result value	xs:double		The result of the analytical measure reported in mg/kg if resType = "VAL".
R.19	resValRec	Result value recovery	xs:double		Recovery value associated with the concentration measurement expressed as a percentage (%). i.e. report 100 for 100%.
R.20	resValRecCorr	Result value corrected for recovery	xs:string (1)	YESNO	Define if the result value has been corrected by calculation for recovery.
R.21	resValUncertSD	Result value uncertainty Standard deviation	xs:double		Standard deviation for the uncertainty measure.
R.22	resValUncert	Result value uncertainty	xs:double		Indicate the expanded uncertainty (usually 95% confidence interval) value associated with the measurement expressed in the unit reported in the field "Result unit".
R.23	moistPerc	Percentage of moisture in the original sample	xs:double		Percentage of moisture in the original sample.
R.24	fatPerc	Percentage of fat in the original sample	xs:double		Percentage of fat in the original sample.
R.25	exprRes	Expression of result	xs:string (5)	EXRES	Code to describe how the result has been expressed: Whole weight, fat weight, dry weight, etc
R.27	resType	Type of result	xs:string (3)	VALTYP	Indicate the type of result, whether it could be quantified/determined or not.
R.28	resLegalLimit	Legal Limit for the result	xs:double		Report the legal limit for the analyte in the product sampled
R.29	resLegalLimitType	Type of legal limit	xs:string(5)	LMTTYP	Type of legal limit applied for the evaluation of the result. ML, MRPL, MRL, action limit etc.
R.30	resEvaluation	Evaluation of the result	xs:string (5)	RESEVAL	Indicate if the result exceeds a legal limit.
R.31	actTakenCode	Action Taken	xs:string (5)	ACTION	Describe any follow-up actions taken as a result of the exceeding a legal limit.
R.32	resComm	Comment of the result	xs:string (250)		Additional comments for this analytical result.

## Appendix E. Glossary

Accuracy	Closeness of agreement between a test result and the true, or the accepted reference value. When applied to a set of test results, it involves a combination of random error (estimated as precision) and a common systematic error (trueness or bias) (ISO 5725-1).
Analyte	The chemical species of which the concentration (or mass) is to be determined. For the purposes of these procedures: a pesticide or a metabolite, breakdown product or derivative of a pesticide or an internal standard.
Analytical sample	See test sample.
Analytical portion	See test portion.
API	Atmospheric pressure ionisation. A generic term including electrospray ionisation (ESI) and atmospheric pressure chemical ionisation (APCI).
AQC	Analytical quality control. Measurement and recording requirements intended to demonstrate the performance of the analytical method in routine practice. The data supplement those generated at method validation. AQC data may be used to validate the extension of methods to new analytes, new matrices and new levels. Synonymous with the terms internal quality control (IQC) and performance verification. Concurrent AQC data are those generated during analysis of the batch in which the particular sample is included.
Batch (analysis)	For extraction, clean-up and similar processes, a batch is a series of samples dealt with by an analyst (or team of analysts) in parallel, usually in one day, and should incorporate at least one recovery determination. For the determination system, a batch is a series undertaken without a significant time break and which incorporates all relevant calibration determinations (also referred to as an "analysis sequence", a "chromatography sequence", etc.). A determination batch may incorporate more than one extraction batch.
	This document does not refer to "batch" in the IUPAC or Codex sense, which relates to manufacturing or agricultural production batches.
Bias	The difference between the mean measured value and the true value, i.e. the total systematic error.
Blank	(i) Material (a sample, or a portion or extract of a sample) known not to contain detectable levels of the analyte(s) sought. Also known as a matrix blank.
	(ii) A complete analysis conducted using the solvents and reagents only; in the absence of any sample (water may be substituted for the sample, to make the analysis realistic). Also known as a reagent blank or procedural blank.
Bracketing calibration	Organisation of a batch of determinations such that the detection system is calibrated immediately before and after the analysis of the samples. For example, calibrant 1, calibrant 2, sample 1, sample $n$ , calibrant 1, calibrant 2.

Calibration	Determination of the relationship between the observed signal (response produced by the detection system) from the target analyte in the sample extract and known quantities of the analyte prepared as standard solutions. In the present document, calibration does not refer to calibration of weighing and volumetric equipment, mass calibration of mass spectrometers, and so on.
Calibration standard	A solution (or other dilution) of the analyte (and internal standard, if used) used for calibration of the determination system. May be prepared from a working standard and may be matrix-matched.
Certified reference material (CRM)	See reference material.
CI	Chemical ionisation for GC-MS and GC-MS/MS.
Comminution	The process of reducing a solid sample to smaller fragments by crushing, pulverising, grinding, etc.
Confirmation	Confirmation is the combination of two or more analyses that are in agreement with each other (ideally, using methods of orthogonal selectivity), at least one of which meets identification criteria) <sup>7</sup> .
	It is impossible to confirm the complete absence of residues. Adoption of a "reporting limit" at the LCL avoids the unjustifiably high cost of confirming the presence, or absence, of residues at unnecessarily low levels.
	The nature and extent of confirmation required for a positive result depends upon importance of the result and the frequency with which similar residues are found.
	Assays based on an ECD tend to demand confirmation, because of their lack of specificity.
	Mass spectrometric techniques are often the most practical and the least equivocal approach to confirmation.
	AQC procedures for confirmation should be rigorous.
Contamination	Unintended introduction of the analyte into a sample, extract, internal standard solution etc., by any route and at any stage during sampling or analysis.
Determination/detection system	Any system used to detect and determine the concentration or mass of the analyte. For example, GC-FPD, LC-MS/MS, LC with post-column derivatisation.
Diagnostic ion	Mass spectrometric term for ions that are highly characteristic for the compound measured.
ECD	Electron-capture detector.
El	Electron ionisation.
EU	European Union.
False negative	A result wrongly indicating that the analyte concentration does not exceed a specified value.
False positive	A result wrongly indicating that the analyte concentration exceeds a specified value.

<sup>&</sup>lt;sup>7</sup>EURACHEM/CITAC Guide "Use of uncertainty information in compliance assessment" (1st Ed., 2007)

FPD & PFPD	Flame-photometric detector and pulsed flame photometric detector (may be specific to sulphur or phosphorus detection).
FWHM	Full-width at half maximum.
GC	Gas chromatography (gas-liquid chromatography).
Identification	Is a qualitative result from a method capable of providing structural information (e.g., using mass spectrometric (MS) detection) that meets acceptable criteria for the purpose of the analysis. The process of generating of sufficient evidence to ensure that a result for a specific sample is valid. Analytes must be identified correctly in order to be quantified.
	AQC procedures for identification should be rigorous.
Interference	A positive or negative response produced by a compound(s) other than the analyte, contributing to the response measured for the analyte, or making integration of the analyte response less certain or accurate. Interference is also loosely referred to as "chemical noise" (as distinct from electronic noise, "flame noise", etc.). Matrix effects are a subtle form of interference. Some forms of interference may be minimised by greater selectivity of the detector. If interference cannot be eliminated or compensated, its effects may be acceptable if there is no significant impact on accuracy.
High resolution MS	Detection using mass spectrometers with high resolving power, typically > 20,000 FWHM .
Internal quality control (IQC)	See AQC.
Within-laboratory reproducibility	See reproducibility.
Internal standards	Definitions are given in the main body of text.
Laboratory sample	The sample sent to and received by the laboratory.
LC	Liquid chromatography (primarily high performance liquid chromatography, HPLC and Ultra high performance liquid chromatography, UPLC).
LCL	Lowest calibrated level. The lowest concentration (or mass) of analyte with which the determination system is successfully calibrated, throughout the analysis batch. See also "reporting limit".
LC-MS	Liquid chromatographic separation coupled with mass spectrometric detection.
Level	In this document, refers to concentration (e.g. mg/kg, µg/ml) or quantity (e.g. ng, pg).
LOD (as referred to in Reg. 396/2005)	Limit of determination (LOD) means the validated lowest residue concentration which can be quantified and reported by routine monitoring with validated control methods; In this respect it can be regarded as the LOQ (see below)
LOQ	Limit of quantitation (quantification). The lowest concentration or mass of the analyte that has been validated with acceptable accuracy by applying the complete analytical method. LOQ is preferable to LOD because it avoids possible confusion with "limit of detection". However, In Reg.396/2005 MRLs that are set at the limit of quantification/determination are referred to as "LOD MRLs", not "LOQ MRLs".

Mass accuracy: Mass resolution	Mass accuracy is the deviation of the measured accurate mass from the calculated exact mass of an ion. It can be expressed as an absolute value in milliDaltons (mDa) or as a relative value in parts-per-million (ppm) error and is calculated as follows: (accurate mass – exact mass) Example: the experimentally measured mass = 239.15098, the theoretical exact mass of the ion m/z = 239.15028. The mass accuracy = (239.15098 – 239.15028) = 0.7 mDa or (accurate mass – exact mass) / exact mass * 10 <sup>6</sup> Example: the experimentally measured mass = 239.15098, the theoretical exact mass of the ion m/z = 239.15028 The mass accuracy = (239.15098 – 239.15028) / 239.15098, the theoretical exact mass of the ion m/z = 239.15028 * 10 <sup>6</sup> = 2.9 ppm The resolution of a mass spectrometry instrument is the ability to
	distinguish between two ions with similar m/z values (IUPAC definition <sup>8,8</sup> : the smallest mass difference between two equal magnitude peaks so that the valley between them is a specified fraction of the peak height).
Mass resolving power	The resolving power, defined at full-width half maximum (FWHM), is g8 m/ $\Delta$ m, where m is the m/z being measured and $\Delta$ m the width of the mass peak at half peak height.
	Note 1: for magnetic sector instruments another definition is used ("10% valley"). Roughly the difference between the two definitions is a factor of 2 (i.e. 10,000 resolving power by the 10% valley method equals 20,000 resolving power by FWHM).
	Note 2: mass resolving power is often confused or interchangeably used with mass resolution (see definition above).
Matrix blank	See blank.
Matrix effect	An influence of one or more undetected components from the sample on the measurement of the analyte concentration or mass. The response of some determination systems (e.g. GC-MS, LC-MS/MS) to certain analytes may be affected by the presence of co-extractives from the sample (matrix). Partition in headspace analyses and SPME is also frequently affected by components present in the samples. These matrix effects derive from various physical and chemical processes and may be difficult or impossible to eliminate. They may be observed as increased or decreased detector responses, compared with those produced by simple solvent solutions of the analyte. The presence, or absence, of such effects may be demonstrated by comparing the response produced from the analyte in a simple solvent solution with that obtained from the same quantity of analyte in the presence of the sample or sample extract. Matrix effects tend to be variable and unpredictable in occurrence, although certain techniques and systems (e.g. HPLC-UV, isotope dilution) are inherently less likely to be influenced. More reliable calibration may be obtained with matrix-matched calibration when it is necessary to use techniques or equipment that are potentially prone to the effects. Matrix-matched calibration may compensate for matrix effects but does not eliminate the underlying cause. Because the underlying cause remains, the intensity of effect may differ from one matrix or sample to another, and also according to the "concentration" of matrix. Isotope dilution or standard addition may be used where matrix effects are sample dependent.

<sup>&</sup>lt;sup>8</sup>On the basis of OECD Environment, Health and safety Publications, Series on Testing and Assessment , No72 and Series on Pesticides No. 39

<sup>&</sup>lt;sup>8</sup> S.J. Lehotay, K. Mastovska, A. Amirav, A.B. Fialkov, T. Alon, P.A. Martos, A. de Kok, A.R. Fernandez-Alba, Trends in Anal. Chem. 27 (2008) 1070-1090

Matrix-matched	Calibration intended to compensate for matrix effects and acceptable
calibration	interference, if present. The matrix blank (see "blank") should be prepared as for analysis of samples. In practice, the pesticide is added to a blank extract (or a blank sample for headspace analysis) of a matrix similar to that analysed. The blank matrix used may differ from that of the samples if it is shown to compensate for the effects. However, for determination of residues approaching or exceeding the MRL, the same matrix (or standard addition) should be used.
Method	A sequence of analytical procedures or steps, from receipt of a sample through to the calculation of results.
Method development	The process of design and preliminary assessment of the characteristics of a method, including ruggedness.
Method validation	The process of characterising the performance to be expected of a method in terms of its scope, specificity, accuracy sensitivity, repeatability and within laboratory reproducibility. Some information on all characteristics, except within laboratory reproducibility, should be established prior to the analysis of samples, whereas data on reproducibility and extensions of scope may be produced from AQC, during the analysis of samples. Wherever possible, the assessment of accuracy should involve analysis of certified reference materials, participation in proficiency tests, or other inter-laboratory comparisons.
MRL	Maximum residue level. In Regulation 396/2005 list MRLs for pesticide/commodity combinations, an asterisk indicates that the MRL* is set at or about the LOQ, with the LOQ being here a consensus figure rather than a measured value.
MS	Mass spectrometry.
MS/MS	Tandem mass spectrometry, here taken to include $MS^n$ . An $MS$ procedure in which ions of a selected mass to charge ratio ( $m/z$ ) from the primary ionisation process are isolated, fragmented usually by collision, and the product ions separated ( $MS/MS$ or $MS^2$ ). In ion-trap mass spectrometers, the procedure may be carried out repetitively on a sequence of product ions ( $MS^n$ ), although this is not usually practical with low-level residues.
Мау	MAY within this document means perhaps or possibly an option (the action is optional).
Must	MUST within this document means an absolute requirement (the action is mandatory).
	MUST NOT means an absolute no.
NPD	Nitrogen-phosphorus detector.
Non-compliance	See violative residue, or MRL exceedance.
Performance verification	see analytical quality control (AQC).
Precision	The closeness of agreement between independent analytical results obtained by applying the experimental procedure under stipulated conditions. The smaller the random part of the experimental errors which affect the results, the more precise the procedure. A measure of precision (or imprecision) is the <u>standard deviation</u> .

Priming (of GC injectors and columns)	Priming effects resemble long-lasting matrix effects and are typically observed in gas chromatography. Typically, an aliquot of sample extract that has not been subjected to clean-up may be injected after a new column or injector liner is fitted, or at the beginning of a batch of determinations. The objective is to "deactivate" the GC system and maximise transmission of the analyte to the detector. In some cases, large quantities of analyte may be injected with the same objective. In such cases it is critically important that injections of solvent or blank extracts are made before samples are analysed, to ensure the absence of carryover of the analyte. Priming effects are rarely permanent and may not eliminate matrix effects.
Procedural blank	See blank.
"Pure" standard	A solid, liquid or gaseous compound that has been prepared in a largely purified form and packed appropriately to ensure stability and allow transportation and storage. The storage conditions, expiry date, purity must be indicated as well as the hydratation water content and the isomer composition where this is relevant.
	Where standards are bought in solution they should be treated as secondary standards (i.e. as stock or working solutions).
(Quasi)-molecular ion	A molecular ion ( $M^+$ or $M^-$ ) or a protonated ( $M+H^+$ ) or deprotonated molecule ( $M-H^+$ ).
Reagent blank	See blank.
Recovery (of analyte through an analytical method)	The proportion of analyte remaining at the point of the final determination, following its addition (usually to a blank sample) immediately prior to extraction. Usually expressed as a percentage.
	Routine recovery refers to the determination(s) performed with the analysis of each batch of samples.
Reference material	Material characterised with respect to its notionally homogeneous content of analyte. Certified reference materials (CRMs) are normally characterised in a number of laboratories, for concentration and homogeneity of distribution of analyte. In-house reference materials are characterised in the owner's laboratory and the accuracy may be unknown.
Reference spectrum	A spectrum of absorption (e.g. UV, IR), fluorescence, ionisation products (MS), etc., derived from the analyte and which may be characteristic of it. The reference mass spectrum preferably should be produced from the "pure" standard (or a solution of the "pure" standard) by the instrument used for analysis of the samples, and similar ionisation conditions must be used.
Repeatability (r)	The precision (standard deviation) of measurement of an analyte (usually obtained from recovery or analysis of reference materials), obtained using the same method on the same sample(s) in a single laboratory over a short period of time, during which differences in the materials and equipment used and/or the analysts involved will not occur. The measure of precision usually is expressed in terms of imprecision and computed as standard deviation of the test result.
	May also be defined as the value below which the absolute difference between two single test results on identical material, obtained under the above conditions, may be expected to lie with a specified probability (e.g. 95%).

Reporting limit (RL)	The lowest level at which residues will be reported as absolute numbers. It is equal to, or higher than the LOQ. For EU monitoring purposes where samples for surveys are analysed over a 12-month period, the same reporting limit should be achievable throughout the whole year.
Representative analyte	An analyte used to assess probable analytical performance in respect of other analytes notionally sought in the analysis. Acceptable data for a representative analyte are assumed to show that performance is satisfactory for the represented analytes. Representative analytes must include those for which the worst performance is expected.
Reproducibility (R)	The precision (standard deviation) of measurement of an analyte (usually by means of recovery or analysis of reference materials), obtained using the same method in a number of laboratories, by different analysts, or over a period in which differences in the materials and equipment will occur. The measure of precision usually is expressed in terms of imprecision and computed as standard deviation of the test result.
	Within-reproducibility (w_R) is that produced in a single laboratory under these conditions.
	May also be defined as the value below which the absolute difference between two single test results on identical material, obtained under the above conditions, may be expected to lie with a specified probability (e.g. 95%).
Response	The absolute or relative signal output from the detector when presented with the analyte.
RSD	Relative standard deviation (coefficient of variation).
Sample	A general term with many meanings but, in these guidelines, refers to laboratory sample, test sample, test portion, or an aliquot of extract.
Sample preparation	The first of two processes which may be required to convert the laboratory sample into the test sample. The removal of parts that are not to be analysed, if required.
Sample processing	The second of two processes which may be required to convert the laboratory sample into the test sample. The process of homogenization, comminution, mixing, etc., if required.
SDL (qualitative screening)	The screening detection limit of a qualitative screening method is the lowest concentration for which it has been demonstrated that a certain analyte can be detected (not necessarily meeting unequivocal identification criteria ) in at least 95% of the samples (i.e. a false-negative rate of 5% is accepted).
SD	Standard deviation.
Selectivity	The ability of the extraction, the clean-up, the derivatisation, the separation system and (especially) the detector to discriminate between the analyte and other compounds. GC-ECD is a selective determination system providing no specificity.

Should	SHOULD within this document means a recommendation that may be ignored but only in particular circumstances (because of valid reasons) and the full implications of ignoring the recommendation must be understood and carefully assessed before choosing a different course of action. SHOULD NOT means not recommended, although it may be acceptable in particular circumstances, but the full implications of ignoring the recommendation must be understood and carefully assessed.
Significant figures	Those digits in a number that are known with certainty, plus the first uncertain digit.
	Ex. 3 significant figures
	0.104, 1.04, 104, 1.04 x10 <sup>4</sup>
	The 1 and the middle 0 are certain, and the 4 is uncertain, but significant.
	Note: Initial zeroes are never significant. Exponential number has no effect on the number of significant figures.
SIM	Selected ion monitoring. Operation of a mass spectrometer in which the abundance of several ions of specific m/z values are recorded rather than the entire mass spectrum
SRM	Selected reaction monitoring. Measurement of specific product ions corresponding to m/z selected precursor ions recorded via two or more stages of mass spectrometry (MS <sup>n</sup> ).
Solid phase dilution	Dilution of a pesticide by distribution within a finely divided solid, such as starch powder. Normally used only for insoluble analytes such as the complex dithiocarbamates.
S/N	Signal-to-noise ratio.
Specificity	The ability of the detector (supported by the selectivity of the extraction, clean-up, derivatisation or separation, if necessary) to provide signals that effectively identify the analyte. GC-MS with El is a fairly non-selective determination system capable of high specificity. High resolution mass MS and MS <sup>n</sup> can be both highly selective and highly specific.
Spike or spiking	Addition of analyte for the purposes of recovery determination or standard addition.
SPME	Solid phase micro-extraction.
Standard	A general term which may refer to a "pure" standard, stock standard, working standard, or calibration standard.
Stock standard solution	The most concentrated solution (or solid dilution, etc.) of the "pure" standard or internal standard, from which aliquots are used to prepare working standard solutions or calibration standard solutions
Test portion	Also referred to as the "analytical portion".
	A representative sub-sample of the test sample, i.e. the portion which is to be analysed.

Test sample	Also referred to as the "analytical sample".
	The laboratory sample after removal of any parts that are not to be analysed, e.g. bones, adhering soil. It may or may not be comminuted and mixed before withdrawing test portions. See also Directive 2002/63/EC.
Trueness	The measure of trueness is normally expressed as 'bias'.
	The closeness of agreement between the average value obtained from a series of test results (i.e. the mean recovery) an accepted reference or true value (ISO 5725-1).
Uncertainty (of measurement)	A range around the reported result within which the true value can be expected to lie with a specified probability (confidence level, usually 95%). Uncertainty data should encompass trueness (bias) and reproducibility
Unit (sample)	A single fruit, vegetable, animal, cereal grain, can, etc. For example, an apple, a T-bone steak, a grain of wheat, a can of tomato soup.
Validation	see method validation
Violative residue	A residue which exceeds the MRL or is unlawful for any other reason.
Working standard solution	A general term used to describe dilutions produced from the stock standard, which are used, for example, to spike for recovery determination or to prepare calibration standard solutions.